

STIC Search Report Biotech-Chem Library

STIC Database Tracking Number: 111228

TO: Hong Liu

Location: CM1/4E01

Art Unit: 1624

Wednesday, January 07, 2004

Case Serial Number: 09/678595

From: Barb O'Bryen

Location: Biotech-Chem Library

CM1-6A05

Phone: 308-4291

barbara.obryen@uspto.gov

Search Notes



THIS PAGE BLANK (USPTO)

=> fil reg; d stat que l16; fil capl; d que nos l18; fil uspatf; d que nos l19; dup rem l18,l19

FILE 'REGISTRY' ENTERED AT 16:13:46 ON 07 JAN 2004

USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT.

PLEASE SEE "HELP USAGETERMS" FOR DETAILS.

COPYRIGHT (C) 2004 American Chemical Society (ACS)

Property values tagged with IC are from the ZIC/VINITI data file provided by InfoChem.

STRUCTURE FILE UPDATES: 5 JAN 2004 HIGHEST RN 634558-38-6 DICTIONARY FILE UPDATES: 5 JAN 2004 HIGHEST RN 634558-38-6

TSCA INFORMATION NOW CURRENT THROUGH JULY 14, 2003

Please note that search-term pricing does apply when conducting SmartSELECT searches.

Crossover limits have been increased. See HELP CROSSOVER for details.

Experimental and calculated property data are now available. For more information enter HELP PROP at an arrow prompt in the file or refer to the file summary sheet on the web at: http://www.cas.org/ONLINE/DBSS/registryss.html

L10	165325	SEA FILE=REGISTRY ABB=ON	16.239/RID = () (all broads
L11	371280	SEA FILE=REGISTRY ABB=ON	16.239/RID = (N) (all bonds 16.165/RID (N) (unspec, Fied
L12	490595	SEA FILE=REGISTRY ABB=ON	46.383/RID (inspecified
L13	567205	SEA FILE=REGISTRY ABB=ON	16.195/RID (3
L15	172	SEA FILE=REGISTRY ABB=ON	PT/ELS AND ((3/N AND (L11 OR L12 OR
		L13)) OR (2/N AND L10))	- compound contains platinum
L16	52	SEA FILE=REGISTRY ABB=ON	L15 AND 1/NR
		•	Lonly Irina
			ر ر

0

FILE 'CAPLUS' ENTERED AT 16:13:46 ON 07 JAN 2004 USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT. PLEASE SEE "HELP USAGETERMS" FOR DETAILS. COPYRIGHT (C) 2004 AMERICAN CHEMICAL SOCIETY (ACS)

Copyright of the articles to which records in this database refer is held by the publishers listed in the PUBLISHER (PB) field (available for records published or updated in Chemical Abstracts after December 26, 1996), unless otherwise indicated in the original publications. The CA Lexicon is the copyrighted intellectual property of the American Chemical Society and is provided to assist you in searching databases on STN. Any dissemination, distribution, copying, or storing of this information, without the prior written consent of CAS, is strictly prohibited.

FILE COVERS 1907 - 7 Jan 2004 VOL 140 ISS 2 FILE LAST UPDATED: 6 Jan 2004 (20040106/ED)

This file contains CAS Registry Numbers for easy and accurate substance identification.

'OBI' IS DEFAULT SEARCH FIELD FOR 'CAPLUS' FILE

L10 165325 SEA FILE=REGISTRY ABB=ON 16.239/RID L11 371280 SEA FILE=REGISTRY ABB=ON 16.165/RID

<<<

FILE 'USPATFULL' ENTERED AT 16:13:46 ON 07 JAN 2004
CA INDEXING COPYRIGHT (C) 2004 AMERICAN CHEMICAL SOCIETY (ACS)

FILE COVERS 1971 TO PATENT PUBLICATION DATE: 6 Jan 2004 (20040106/PD)
FILE LAST UPDATED: 6 Jan 2004 (20040106/ED)
HIGHEST GRANTED PATENT NUMBER: US6675388
HIGHEST APPLICATION PUBLICATION NUMBER: US2004003444
CA INDEXING IS CURRENT THROUGH 6 Jan 2004 (20040106/UPCA)
ISSUE CLASS FIELDS (/INCL) CURRENT THROUGH: 6 Jan 2004 (20040106/PD)
REVISED CLASS FIELDS (/NCL) LAST RELOADED: Oct 2003
USPTO MANUAL OF CLASSIFICATIONS THESAURUS ISSUE DATE: Oct 2003

```
USPAT2 is now available. USPATFULL contains full text of the
                                                                        <<<
>>>
    original, i.e., the earliest published granted patents or
                                                                        <<<
>>>
    applications. USPAT2 contains full text of the latest US
                                                                        <<<
    publications, starting in 2001, for the inventions covered in
                                                                        <<<
    USPATFULL. A USPATFULL record contains not only the original
                                                                        <<<
    published document but also a list of any subsequent
                                                                        <<<
>>>
    publications. The publication number, patent kind code, and
                                                                        <<<
    publication date for all the US publications for an invention
                                                                        <<<
>>>
    are displayed in the PI (Patent Information) field of USPATFULL
                                                                        <<<
>>>
    records and may be searched in standard search fields, e.g., /PN,
>>>
    /PK, etc.
                                                                        <<<
>>>
     USPATFULL and USPAT2 can be accessed and searched together
                                                                        <<<
    through the new cluster USPATALL. Type FILE USPATALL to
                                                                        <<<
    enter this cluster.
>>>
                                                                        <<<
>>>
                                                                        <<<
     Use USPATALL when searching terms such as patent assignees,
>>>
                                                                        <<<
>>>
     classifications, or claims, that may potentially change from
                                                                        <<<
```

This file contains CAS Registry Numbers for easy and accurate substance identification.

the earliest to the latest publication.

FILE 'CAPLUS' ENTERED AT 16:13:46 ON 07 JAN 2004 USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT. PLEASE SEE "HELP USAGETERMS" FOR DETAILS. COPYRIGHT (C) 2004 AMERICAN CHEMICAL SOCIETY (ACS)

FILE 'USPATFULL' ENTERED AT 16:13:46 ON 07 JAN 2004
CA INDEXING COPYRIGHT (C) 2004 AMERICAN CHEMICAL SOCIETY (ACS)

```
PROCESSING COMPLETED FOR L18
PROCESSING COMPLETED FOR L19
L21
```

26 DUP REM L18 L19 (0 DUPLICATES REMOVED) ANSWERS '1-25' FROM FILE CAPLUS ANSWER '26' FROM FILE USPATFULL

=> d ibib abs hitstr 1-26; fil cao; d que nos 120

ANSWER 1 OF 26 CAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER: 2003:173417 CAPLUS

DOCUMENT NUMBER: 138:214473

Platinum complexes containing nonplanar heterocyclic TITLE:

aliph. amines and their use in cancer treatment

Barenholz, Yechezkel; Gibson, Dan; Najajreh, Yousef; INVENTOR(S):

Khazanov, Elena

Yissum Research Development Company of the Hebrew PATENT ASSIGNEE(S):

University of Jerusalem, Israel

SOURCE: PCT Int. Appl., 67 pp.

CODEN: PIXXD2

DOCUMENT TYPE:

Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

```
PATENT NO.
                KIND DATE
                                   APPLICATION NO.
                                                      DATE
                                     -----
                ----
WO 2003017998
                      20030306
                                     WO 2002-IL687
                                                      20020821
                 Α1
                      20030828
WO 2003017998
                 В1
       AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN,
       CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH,
       GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR,
       LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH,
        PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TN, TR, TT, TZ,
        UA, UG, US, UZ, VC, VN, YU, ZA, ZM, AW, AZ, BY, KG, KZ, MD,
        RU, TJ, TM
   RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AT, BE, BG,
       CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL,
        PT, SE, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR,
        NE, SN, TD, TG
```

PRIORITY APPLN. INFO.:

US 2001-314115P P 20010823

OTHER SOURCE(S): MARPAT 138:214473

The present invention concerns novel Pt complexes in which at least one of the amine ligands is a nonplanar heterocyclic aliph. amine. Claimed are platinum complexes [Pt(X)(Y)(Am1)(Am2)] [X and Y = halo, carboxylate, phosphate, or sulfate; Am1 = amine selected from NH3, primary amine, secondary amine, non-planar heterocyclic aliph. amine or heterocyclic arom. amine; Am2 = non-planar heterocyclic aliph. amine, with the proviso that when said complex has cis configuration, Am1 and Am2 cannot represent simultaneously piperidine] and their pharmaceutical compns. Prepd. claimed complexes include trans-[PtCl2(NH3)(Am2)] (Am2 = piperidine, 4-hydroxypiperidine, 4-piperidinopiperidine, etc.), cis-[PtCl2(NH3)(piperidine)] or cis-[PtCl2(NH3)(piperazine)].cntdot.HCl, and linked dinuclear complex bis[{trans,trans-PtCl2(piperazine)2}{4,7,10trioxo-1,13-tridecanediamine}].cntdot.2HCl. The Pt complexes may be in a trans or cis configuration and possess therapeutic antitumor activities. Achieving a therapeutic effect may comprise forming an adduct between the Pt complex and DNA, inducing apoptosis of undesired cell proliferation, and loading the Pt complex into a liposome. Cytotoxicity assays revealed that replacing one or both ammines of transplatin enhanced significantly the cytotoxicity of the trans-PtCl2 compds. in both human C-26 colon and human OV-1063 ovarian cancer cell lines. Improved antitumor activity was detd. for piperidine- and piperazine-contg. platinum complexes against resistance cancer cell lines A2780/A2780cisR, 41M/41McisR, and

CH1/CH1cisR.

IT 499769-13-0P 499769-14-1P

RL: SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES (Uses)

(prepn. as antitumor agent)

RN 499769-13-0 CAPLUS

CN Platinum, dichloro(1-nonanamine)(piperazine-.kappa.N1)-, monohydrochloride, (SP-4-1)- (9CI) (CA INDEX NAME)

$$\begin{array}{c|c} \text{C1}^-\\ & 2+\\ \text{Pt} & \text{NH}_2-\text{ (CH}_2)_8-\text{Me} \\ & \\ & \text{C1}^- \end{array}$$

● HCl

RN 499769-14-1 CAPLUS

CN Platinum, amminedichloro(1-piperazineethanol-.kappa.N4)-, monohydrochloride, (SP-4-1)- (9CI) (CA INDEX NAME)

HC1

IT 499982-35-3P

RL: BSU (Biological study, unclassified); SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES (Uses)

(prepn. as antitumor agent and DNA binding)

RN 499982-35-3 CAPLUS

CN Platinum, amminedichloro(piperazine-.kappa.N1)-, (SP-4-3)- (9CI) (CA INDEX NAME)

IT 478844-04-1P

RL: PAC (Pharmacological activity); SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES (Uses)

(prepn. as antitumor agent and activity against C-26 colon and OV-1063 ovarian cancer cell lines and resistance cancer cell lines A2780/A2780cisR, 41M/41McisR, and CH1/CH1cisR)

RN 478844-04-1 CAPLUS

CN Platinum, amminedichloro(piperazine-.kappa.N1)-, monohydrochloride, (SP-4-2)- (9CI) (CA INDEX NAME)

HC1

IT 478842-92-1P 478843-07-1P

RL: PAC (Pharmacological activity); SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES (Uses)

(prepn. as antitumor agent and activity against resistance cancer cell lines A2780/A2780cisR, 41M/41McisR, and CH1/CH1cisR)

RN 478842-92-1 CAPLUS

CN Platinum, (1-butanamine)dichloro(piperazine-.kappa.N1)-, monohydrochloride, (SP-4-1)- (9CI) (CA INDEX NAME)

HCl

RN 478843-07-1 CAPLUS

CN Platinum, dichloro(piperazine-.kappa.N1)(2-propanamine)-, monohydrochloride, (SP-4-1)- (9CI) (CA INDEX NAME)

HC1

IT 478842-82-9P

RL: BSU (Biological study, unclassified); PAC (Pharmacological activity); SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES (Uses)

(prepn. as antitumor agent, activity against resistance cancer cell lines A2780/A2780cisR, 41M/41McisR, and CH1/CH1cisR, and DNA binding)

RN 478842-82-9 CAPLUS

Platinum, amminedichloro(piperazine-.kappa.N1)-, monohydrochloride, (SP-4-1)- (9CI) (CA INDEX NAME)

CN

● HCl

REFERENCE COUNT:

9 THERE ARE 9 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

LANSWER 2 OF 26 CAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER: 2003:326574 CAPLUS

DOCUMENT NUMBER: 139:62614

TITLE: DNA Binding Mode of the Cis and Trans Geometries of

New Antitumor Nonclassical Platinum Complexes Containing Piperidine, Piperazine, or 4-Picoline

Ligand in Cell-Free Media. Relations to Their Activity

in Cancer Cell Lines

AUTHOR(S): Kasparkova, Jana; Marini, Victoria; Najajreh, Yousef;

Gibson, Dan; Brabec, Viktor

CORPORATE SOURCE: Institute of Biophysics, Academy of Sciences of the

Czech Republic, Brno, CZ-61265, Czech Rep.

SOURCE: Biochemistry (2003), 42(20), 6321-6332

CODEN: BICHAW; ISSN: 0006-2960

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal LANGUAGE: English

The global modification of mammalian and plasmid DNAs by novel platinum compds., cis- or trans-[PtCl2(NH3)(Am)], where Am = NH3, nonplanar heterocycle piperidine, piperazine, or arom. planar heterocycle 4-picoline, was investigated in cell-free media using various biochem. and biophys. methods. These modifications have been compared with the activity of these new compds. in several tumor cell lines including those resistant to antitumor cis-diamminedichloroplatinum(II) (cisplatin). The results show that the replacement of the NH3 group in cisplatin by the heterocyclic ligands does not considerably affect the DNA binding mode of this drug. Cytotoxicity studies have revealed that the replacement lowers the activity of the platinum compd. in both sensitive and resistant cell lines. It has been suggested that the reduced activity of these analogs of cisplatin is assocd. with some features of the damaged DNA and/or its cellular processing. Alternatively, the reduced activity of the analogs of cisplatin might also be due to the factors that do not operate directly at the level of the target DNA, such as intracellular platinum uptake. In contrast to the analogs of cisplatin, the replacement of one ammine group by the heterocyclic ligand in its clin. ineffective trans isomer (transplatin) results in a radical enhancement of its activity in tumor

cell lines. Importantly, this replacement also markedly alters the DNA binding mode of transplatin. The results support the view that one strategy of how to activate the trans geometry in bifunctional platinum(II) compds. including circumvention of resistance to cisplatin may consist of a chem. modification of the ineffective transplatin that results in an increased stability of its intrastrand cross-links in double-helical DNA and/or in an increased efficiency to form interstrand cross-links.

ΙT 499982-35-3 548446-31-7

> RL: DMA (Drug mechanism of action); PAC (Pharmacological activity); THU (Therapeutic use); BIOL (Biological study); USES (Uses)

(DNA binding mode of cis and trans antitumor nonclassical platinum complexes contg. piperidine and piperazine and 4-picoline ligands in relation to antitumor activity and resistance)

RN 499982-35-3 CAPLUS

CN Platinum, amminedichloro(piperazine-.kappa.N1)-, (SP-4-3)- (9CI)

RN 548446-31-7 CAPLUS

Platinum, amminedichloro(piperazine-.kappa.N1)-, (SP-4-1)- (9CI) INDEX NAME)

REFERENCE COUNT:

49 THERE ARE 49 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

ANSWER 3 OF 26 CAPLUS COPYRIGHT 2004 ACS on STN .

AdCESSION NUMBER:

2002:913261 CAPLUS

DOCUMENT NUMBER:

138:364333

TITLE:

Ligand effects on the binding of cis- and

trans-[PtCl2Am1Am2] to proteins

AUTHOR(S):

Najajreh, Yousef; Peleg-Shulman, Tal; Moshel, Ofra;

Farrell, Nicholas; Gibson, Dan

CORPORATE SOURCE:

School of Pharmacy, Department of Medicinal Chemistry

and Natural Products, The Hebrew University of

Jerusalem, Jerusalem, 91120, Israel

SOURCE:

JBIC, Journal of Biological Inorganic Chemistry

(2003), 8(1-2), 167-175

CODEN: JJBCFA; ISSN: 0949-8257

PUBLISHER:

Springer-Verlag

DOCUMENT TYPE:

Journal

LANGUAGE:

English

As part of a systematic study of the basic principles that govern the formation and reactivity of Pt-protein adducts, we report the effect of substituting the ammine ligand of cis- and trans-[PtCl2(NH3)2] complexes with bulkier planar arom. or nonplanar cyclic amine ligands on the binding

properties of the complexes to ubiquitin and to horse heart myoglobin. The ligand replacement had a different effect on the cis or trans isomers investigated. In the cis-Pt complexes, replacing one or both ammine ligands by piperidine or 4-picoline dramatically decreased the binding of the complexes to the proteins studied, whereas in the substituted trans-Pt complexes replacement of the ammine by a piperidine or 4-picoline increased the binding rate. This behavior may have to do with the different preferred binding sites of the cis- and trans-Pt complexes. bulkier cis- or trans-Pt complexes investigated also did not display a preference for Metl of ubiquitin, possibly owing to steric constraints imposed by the substituted ligands. The introduction of a charged piperazine ligand significantly decreased the rate of binding to the protein, possibly owing to electrostatic interactions or hydrogen-bond formations with the surface of the protein. The binding of the complexes to ubiquitin and myoglobin does not disrupt the folding of the proteins as judged by electrospray ionization mass spectrometry.

IT 478842-82-9P

RN CN RL: BSU (Biological study, unclassified); CPS (Chemical process); PEP (Physical, engineering or chemical process); RCT (Reactant); SPN (Synthetic preparation); BIOL (Biological study); PREP (Preparation); PROC (Process); RACT (Reactant or reagent)

(ligand effects on binding of cis- and trans-[PtCl2Am1Am2] to proteins) 478842-82-9 CAPLUS

Platinum, amminedichloro(piperazine-.kappa.N1)-, monohydrochloride, (SP-4-1)-(9CI) (CA INDEX NAME)

HCl

23 REFERENCE COUNT: THERE ARE 23 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

CAPLUS COPYRIGHT 2004 ACS on STN L21 ANSWER 4 OF 26

ACCESSION NUMBER: 2002:275998 CAPLUS

DOCUMENT NUMBER: 136:288219

TITLE: Platinum ammine complexes or derivatives with improved

aqueous solubility and activity as antitumor agents

INVENTOR(S): Wong, Ernest S. Y.; Giandomenico, Christen M.

PATENT ASSIGNEE(S): Anormed Inc., Can. SOURCE: PCT Int. Appl., 25 pp.

CODEN: PIXXD2

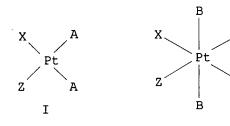
DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.			KIND		DATE			A.	PPLI	CATI	N NC	ο.	DATE				
WO 2002028871			A1 20020411				W	200	01-U	38	20011002						
W:	ΑE,	AG,	AL,	ΑM,	ΑT,	ΑU,	ΑZ,	BA,	BB,	BG,	BR,	BY,	ΒZ,	CA,	CH,	CN,	
	CO,	CR,	CU,	CZ,	DE,	DK,	DM,	DZ,	EC,	EE,	ES,	FI,	GB,	GD,	GE,	GH,	
	GM,	HR,	ΗU,	ID,	IL,	IN,	IS,	JP,	KE,	KG,	ΚP,	KR,	ΚZ,	LC,	LK,	LR,	

```
LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PH, PL,
             PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG,
             US, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM
         RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY,
             DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR, BF,
             BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG
     AU 2001094969
                       Α5
                            20020415
                                           AU 2001-94969
                                                             20011002
                            20030702
                                                             20011002
                                           EP 2001-975671
     EP 1322654
                       Α1
             AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
             IE, SI, LT, LV, FI, RO, MK, CY, AL, TR
PRIORITY APPLN. INFO.:
                                        US 2000-678595
                                                          Α
                                                             20001004
                                        WO 2001-US30838 W
                                                             20011002
GΙ
```



The present invention relates to Pt antitumor drugs. In particular, it relates to Pt complexes I or II or a pharmaceutically acceptable salt thereof wherein each A is independently an anion, each B is independently halo, hydroxy, carboxylate, carbamate or a carbonate ester, Z is a substituted 5- or 6-membered, heterocyclic moiety wherein at least one substituent sterically hinders access of the Pt atom to a DNA strand of a tumor cell, and wherein Z is other than pyridine, and X is NH3 or mono- or dialkyl-substituted NH3, which are active against human cancer cells and have improved aq. soly. and activity. Example complexes which are prepd. include cis-[PtII(NH3)Cl2(L)] (L = 3,5-dimethylpyrazole, 1-methylimidazole, 3,5-dimethylisoxazole, 2,3-dimethylpyrazine, etc.) or (OC-6-43)-[PtIV(NH3)Cl2(OH)2(L)] (L = 3,5-dimethylpyrazole, 1-methylimidazole, 2,3-dimethylpyrazine) or (OC-6-43)-[PtIV(NH3)C12(OAc)2(L)] (L = 2,3-dimethylpyrazine). The aq. soly. of the complexes at ambient conditions are greater than that of cisplatin. activity of the complexes in inhibiting human cell lines is comparable to that of prior art compds., at least in some cell lines. Resistance factors with respect to 41M/41MR are particularly favorable for some of the complexes. ΙT

II

406161-68-0P 406161-69-1P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(intermediate; for prepn. of amminedichloro(N-heterocycle)platinum(II) complex)

406161-68-0 CAPLUS

RN

Platinum, amminechloro(3,5-dimethyl-1H-pyrazole-.kappa.N2)iodo- (9CI) CN INDEX NAME)

RN 406161-69-1 CAPLUS

CN Platinum, amminechloroiodo(1,3,5-trimethyl-1H-pyrazole-.kappa.N2)- (9CI) (CA INDEX NAME)

TT 301299-27-4P 406161-57-7P 406161-60-2P

406161-62-4P 406161-63-5P 406161-64-6P 406161-65-7P 406161-66-8P 406161-67-9P

RL: BSU (Biological study, unclassified); PAC (Pharmacological activity); PRP (Properties); SPN (Synthetic preparation); BIOL (Biological study); PREP (Preparation)

(prepn., aq. soly., and antitumor activity)

RN 301299-27-4 CAPLUS

CN Platinum, amminedichloro(2,5-dimethylpyrazine-.kappa.N1)-, (SP-4-3)- (9CI) (CA INDEX NAME)

RN 406161-57-7 CAPLUS

CN Platinum, amminedichloro(1,3,5-trimethyl-1H-pyrazole-.kappa.N2)-, (SP-4-3)- (9CI) (CA INDEX NAME)

RN 406161-60-2 CAPLUS

CN Platinum, amminedichloro(2-methyl-1H-imidazole-.kappa.N3)-, (SP-4-3)- (9CI) (CA INDEX NAME)

RN 406161-62-4 CAPLUS

CN Platinum, amminedichloro(trimethyloxazole-.kappa.N3)-, (SP-4-3)- (9CI) (CA INDEX NAME)

RN 406161-63-5 CAPLUS

CN Platinum, amminedichloro(3,5-dimethyl-lH-pyrazole-.kappa.N2)dihydroxy-, (OC-6-43)- (9CI) (CA INDEX NAME)

RN 406161-64-6 CAPLUS

CN Platinum, amminedichlorodihydroxy(1-methyl-1H-imidazole-.kappa.N3)-, (OC-6-43)- (9CI) (CA INDEX NAME)

RN 406161-65-7 CAPLUS

CN Platinum, bis(acetato-.kappa.O)amminedichloro(2,3-dimethylpyrazine-.kappa.N1)-, (OC-6-43)- (9CI) (CA INDEX NAME)

RN 406161-66-8 CAPLUS

CN Platinum, amminedichloro(1,2-dimethyl-1H-imidazole-.kappa.N3)dihydroxy-, (OC-6-43)- (9CI) (CA INDEX NAME)

RN 406161-67-9 CAPLUS

CN Platinum, amminedichloro(2,5-dimethyl-1H-imidazole-.kappa.N3)dihydroxy-, (OC-6-43)- (9CI) (CA INDEX NAME)

Me Me NH3
$$-_{HO} -_{Pt} -_{C1} -_{C1} -_{C1}$$

IT 301299-34-3P

RL: PAC (Pharmacological activity); RCT (Reactant); SPN (Synthetic preparation); BIOL (Biological study); PREP (Preparation); RACT (Reactant or reagent)

(prepn., conversion to diacetato deriv., and use as antitumor agent)

RN 301299-34-3 CAPLUS

CN Platinum, amminedichloro(2,3-dimethylpyrazine-.kappa.N1)dihydroxy-, (OC-6-43)- (9CI) (CA INDEX NAME)

TT 114487-38-6P 301299-38-7P 406161-56-6P 406161-59-9P 406161-61-3P

RL: BSU (Biological study, unclassified); PAC (Pharmacological activity); PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); BIOL (Biological study); PREP (Preparation); RACT (Reactant or reagent) (prepn., peroxide oxidn. to dihydroxo deriv., aq. soly., and antitumor activity)

RN 114487-38-6 CAPLUS

CN Platinum, amminedichloro(1-methyl-1H-imidazole-.kappa.N3)-, (SP-4-3)- (9CI) (CA INDEX NAME)

RN 301299-38-7 CAPLUS

CN Platinum, amminedichloro(2,3-dimethylpyrazine-.kappa.N1)-, (SP-4-3)- (9CI) (CA INDEX NAME)

RN 406161-56-6 CAPLUS

CN Platinum, amminedichloro(3,5-dimethyl-1H-pyrazole-.kappa.N2)-, (SP-4-3)- (9CI) (CA INDEX NAME)

RN 406161-59-9 CAPLUS

CN Platinum, amminedichloro(1,2-dimethyl-1H-imidazole-.kappa.N3)-, (SP-4-3)- (9CI) (CA INDEX NAME)

RN 406161-61-3 CAPLUS

CN Platinum, amminedichloro(2,5-dimethyl-1H-imidazole-.kappa.N3)-, (SP-4-3)- (9CI) (CA INDEX NAME)

Me Me N Me
$$-C1-Pt \frac{2+}{N}C1-\frac{1}{N}$$

REFERENCE COUNT:

THERE ARE 14 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

221 ANSWER 5 OF 26 CAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER: 2002:830840 CAPLUS

DOCUMENT NUMBER: 138:32969

TITLE: Novel Soluble Cationic trans-

Diaminedichloroplatinum(II) Complexes that Are Active against Cisplatin Resistant Ovarian Cancer Cell Lines

AUTHOR(S): Najajreh, Yousef; Perez, Jose Manuel;

Navarro-Ranninger, Carmen; Gibson, Dan

CORPORATE SOURCE: Department of Medicinal Chemistry and Natural

Products, School of Pharmacy, The Hebrew University of

Jerusalem, Jerusalem, 91120, Israel

SOURCE: Journal of Medicinal Chemistry (2002), 45(24),

5189-5195

CODEN: JMCMAR; ISSN: 0022-2623

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal LANGUAGE: English

AB Pos. charged, water sol. cis/trans-[PtCl2(piperazine) (Aml)] (where Aml = NH3, n-butylamine, isopropylamine, 4-picoline, piperidine, and piperazine) has significant cytotoxic activity against cisplatin resistant ovarian cancer cells. The charged complexes are taken up by cancer cells much more rapidly than cisplatin and bind to cellular DNA and to calf thymus DNA much faster than cisplatin or transplatin. The platinum-piperazine complexes bind proteins (ubiquitin and myoglobin) very slowly as compared to cisplatin and to their neutral piperidine analogs. Altogether, the results reported here suggest that combination of pos. charged ligands with a trans-Pt(II)Cl center may lead to the discovery of platinum complexes that are able to circumvent cisplatin resistance.

IT 478842-82-9P 478842-92-1P 478843-07-1P 478844-04-1P

RL: PAC (Pharmacological activity); SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES (Uses)

(antitumor aminedichloroplatinum complexés active against cisplatin resistant ovarian cancer cell lines)

RN 478842-82-9 CAPLUS

Platinum, amminedichloro(piperazine-.kappa.N1)-, monohydrochloride, (SP-4-1)- (9CI) (CA INDEX NAME)

CN

HC1

RN 478842-92-1 CAPLUS

N Platinum, (1-butanamine)dichloro(piperazine-.kappa.N1)-, monohydrochloride, (SP-4-1)- (9CI) (CA INDEX NAME)

HC1

RN 478843-07-1 CAPLUS

CN

Platinum, dichloro(piperazine-.kappa.N1)(2-propanamine)-, monohydrochloride, (SP-4-1) - (9CI) (CA INDEX NAME)

$$\begin{array}{c|c} & \text{Cl}^-\\ & 2+\\ \text{Pt} & \text{NH}_2-\text{Pr-i}\\ & \\ & \text{N} & \\ & \text{Cl}^- \end{array}$$

HC1

RN 478844-04-1 CAPLUS

CN Platinum, amminedichloro(piperazine-.kappa.N1)-, monohydrochloride, (SP-4-2)-(9CI)(CA INDEX NAME)

HC1

REFERENCE COUNT: 27

THERE ARE 27 CITED REFERENCES AVAILABLE FOR THIS . RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

CAPLUS COPYRIGHT 2004 ACS on STN L21 ANSWER 6 OF 26

ACCESSION NUMBER:

2001:918107 CAPLUS

DOCUMENT NUMBER:

136:200291

TITLE:

Preferential C-Binding versus N-Binding in Imidazole

Depends on the Metal Fragment Involved

AUTHOR(S): CORPORATE SOURCE: Sini, Gjergji; Eisenstein, Odile; Crabtree, Robert H. Universite de Cergy-Pontoise, Cergy-Pontoise, 95031,

SOURCE:

Inorganic Chemistry (2002), 41(3), 602-604

CODEN: INOCAJ; ISSN: 0020-1669

PUBLISHER:

American Chemical Society

DOCUMENT TYPE:

Journal

LANGUAGE:

English

The relative stability of metal fragment involved imidazole C- and N-bound isomers is discussed. The position of equil., where N-bound imidazole rearranges to the C-bound form, with the proton originally bound to the 2-carbon moving to nitrogen, is discussed. C-bound imidazoles are predicted to be thermodynamically more stable than the conventional N-bound forms for several second- and third-row transition metals.

ΙT 400900-34-7 401815-12-1

RL: PRP (Properties)

(relative energy; preferential C-binding vs. N-binding in imidazole depends on metal fragment involved)

RN 400900-34-7 CAPLUS

Platinum, amminedichloro(1,3-dihydro-2H-imidazol-2-ylidene)-, (SP-4-1)-CN (CA INDEX NAME)

RN 401815-12-1 CAPLUS

CN Platinum, amminedichloro(1H-imidazole-.kappa.N3)-, (SP-4-1)- (9CI) INDEX NAME)

REFERENCE COUNT:

25 THERE ARE 25 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L21 ANSWER 7 OF 26 CAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

2000:742103 CAPLUS

DOCUMENT NUMBER:

133:304927

TITLE: INVENTOR(S): Process for preparing amine platinum complexes Wong, Ernest S. Y.; Giandomenico, Christen M.

PATENT ASSIGNEE(S):

Anormed, Inc., Can. PCT Int. Appl., 45 pp.

SOURCE:

DOCUMENT TYPE:

CODEN: PIXXD2

LANGUAGE:

Patent English

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.			KIND DATE			A	PPLI	CATI	0.	DATE								
								_										
	WO	WO 2000061590			A1 20001019				W	200	00-C	A385		2000	0411			
		W:	ΑE,	AG,	AL,	AM,	AT,	ΑU,	ΑZ,	BA,	BB,	BG,	BR,	BY,	CA,	CH,	CN,	CR,
			CU,	CZ,	DE,	DK,	DM,	DZ,	EE,	ES,	FI,	GB,	GD,	GE,	GH,	GM,	HR,	HU,
			ID,	IL,	IN,	IS,	JP,	ΚE,	KG,	KP,	KR,	KZ,	LC,	LK,	LR,	LS,	LT,	LU,

```
LV, MA, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE,
             SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, UZ, VN, YU, ZA, ZW,
             AM, AZ, BY, KG, KZ, MD, RU, TJ, TM
         RW: GH, GM, KE, LS, MW, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY, DE,
             DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF,
             CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG
     BR 2000009780
                             20020102
                                            BR 2000-9780
                       Α
                                                              20000411
     EP 1165576
                             20020102
                                            EP 2000-918620
                       Α1
                                                              20000411
             AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
             IE, SI, LT, LV, FI, RO
     JP 2002541263
                       T2
                             20021203
                                            JP 2000-610861
                                                              20000411
                                            EE 2001-536
     EE 200100536
                       Α
                             20030217
                                                              20000411
     ZA 2001007965
                       Α
                             20030102
                                            ZA 2001-7965
                                                              20010927
     NO 2001004957
                       Α
                             20011203
                                            NO 2001-4957
                                                              20011012
     BG 106090
                       Α
                             20020628
                                            BG 2001-106090
                                                              20011108
PRIORITY APPLN. INFO.:
                                         US 1999-128939P
                                                          Ρ
                                                              19990413
                                         WO 2000-CA385
                                                           W
                                                              20000411
                         MARPAT 133:304927
```

OTHER SOURCE(S):

The present invention relates to the area of Pt amine drugs. In particular, it relates to an improved process for prepg. Pt complexes PtA2LL1 (Ia) or PtA2Y2LL1 (Ib), comprising: (1a) a 1st step, wherein [PtA4]2-, preferably PtCl42-, is reacted with L under appropriate : conditions in a 1st solvent to form [PtA3(L)]-; (1b) a 2nd step, wherein [PtA3(L)]- is reacted with L' under appropriate conditions in a 2nd solvent to form cis-[PtA2(L')(L)]; (1c) in the case there Y is halogen or hydroxy a third step, wherein cis-[PtA2(L')(L)] is reacted with H2O2, Y2 or halogen contg. oxidant to form c,t,c-[PtA2Y2(L')(L)]; in the case where Y is carboxylate, carbamate or carbonate ester a 4th step, wherein an intermediate, where Y is hydroxy formed in step (1c), is functionalized with an appropriate acylating agent; and (1d) in the case where A is not a halide or is different from the original halide, addnl. step(s) in which the original halide A of an intermediate formed in step 1a or 1b, 1c or 1d is converted to a different halide or a new leaving group(s) A such as mono-dentate hydroxy, alkoxy, carboxylate or bidentate carboxylate, phosphonocarboxylate, diphosphonate, or sulfate; wherein L = amine or NH3, L' = amine but not NH3 and Y is a halogen, hydroxide, carboxylate, carbamate or carbonate ester. For example, K2[PtCl4] in N-methylpyrrolidinone reacted with 2-picoline (pic) to give K[PtCl3L] which in aq. soln. in presence of KCl reacted with NH4OAc in presence of NH4OH to give [PtCl2(NH3)(pic)]. [PtCl2(NH3)(pic)] was oxidized by H2O2 to give to give cis, trans, cis-[PtCl2(OH)2(NH3)(pic)] which was converted to [PtCl(OH)3(NH3)(pic)] and subsequently to [PtCl(OAc)3(NH3)(pic)].

ΙT 301299-27-4P 301299-34-3P

> RL: SPN (Synthetic preparation); PREP (Preparation) (improved prepn. of antitumor agent)

RN 301299-27-4 CAPLUS

CN Platinum, amminedichloro(2,5-dimethylpyrazine-.kappa.N1)-, (SP-4-3)- (9CI) (CA INDEX NAME)

RN 301299-34-3 CAPLUS

CN Platinum, amminedichloro(2,3-dimethylpyrazine-.kappa.N1)dihydroxy-, (OC-6-43)-(9CI) (CA INDEX NAME)

IT 301299-38-7

RL: RCT (Reactant); RACT (Reactant or reagent)

(reactant for improved prepn. of platinum amine complexes as antitumor agents)

RN 301299-38-7 CAPLUS

CN Platinum, amminedichloro(2,3-dimethylpyrazine-.kappa.N1)-, (SP-4-3)- (9CI) (CA INDEX NAME)

REFERENCE COUNT:

THERE ARE 6 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

1 ANSWER 8 OF 26 CAPLUS COPYRIGHT 2004 ACS on STN

6

ACCESSION NUMBER:

1998:558981 CAPLUS

DOCUMENT NUMBER:

129:297999

TITLE:

In vitro and in vivo activity of two Pt(IV) salts

against Leishmania donovani

AUTHOR(S):

Mesa-Valle, C. M.; Rodriguez-Cabezas, M. N.;

Moraleda-Lindez, V.; Craciunescu, D.; Sanchez-Moreno,

M.; Osuna, A.

CORPORATE SOURCE:

Departamento Biologia Aplicada, Facultad Ciencias Experimentales, Universidad Almeria, Almeria, Spain

Pharmacology (1998), 57(3), 160-172

CODEN: PHMGBN; ISSN: 0031-7012

PUBLISHER: S. Karger AG

DOCUMENT TYPE:

Journal

LANCHACE.

English

LANGUAGE:

SOURCE:

The activities of 8 Pt drug complex salts were detd. against L. donovani promastigotes. The 3 most active salts were selected: [PtIVBr6]H2

(pentamidine), [PtIVBr6]H2 (stilbamidine), and [PtIVCl6]H2

(2-piperazinyl(1) Et amine), which induced growth-inhibition rates of >50% at 24 h of treatment and at the max. dosage tested. The cytotoxicity assays on the macrophage cell line J-774 showed high cytotoxicity for the salt [PtIVBr6]H2 (stilbamidine) with a percentage of specific 51Cr release of 58.2% at 24 h of incubation and 100 .mu.g/mL. Meanwhile, assays of the other compds. showed practically no cytotoxicity. The salt [PtIVBr6]H2 (pentamidine) notably inhibited the incorporation of 3H-thymidine in the treated parasites. The ultrastructural alterations obsd. in the flagellates treated with the salts [PtIVCl6]H2 (2-piperazinyl(1)Et amine)

and [PtIVBr6]H2 (pentamidine) suggest that both act preferentially at the nuclear level and at the kinetoplast-mitochondrion complex. Both compds. showed a high in vivo activity in parasitized Wistar rats.

IT 66098-32-6

RL: BAC (Biological activity or effector, except adverse); BSU (Biological study, unclassified); THU (Therapeutic use); BIOL (Biological study); USES (Uses)

(antiparasitic activity of Pt(IV) salts)

RN 66098-32-6 CAPLUS

Platinate(2-), hexachloro-, (OC-6-11)-, dihydrogen, compd. with 1-piperazineethanamine (1:1) (9CI) (CA INDEX NAME)

CM 1

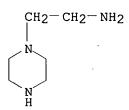
CN

CRN 16941-12-1 CMF C16 Pt . 2 H CCI CCS

2 H+

CM 2

CRN 140-31-8 CMF C6 H15 N3



1/21) ANSWER 9 OF 26 CAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

1995:254588 CAPLUS 122:70654

DOCUMENT NUMBER: TITLE:

Monohistidine complexes of platinum(II) and

palladium(II)

AUTHOR(S): CORPORATE SOURCE:

Krylova, L. F.; Dikanskaya, L. D.; Fedotov, M. A.

Novosib. Gos. Univ., Russia

SOURCE:

Koordinatsionnaya Khimiya (1994), 20(10), 780-5

CODEN: KOKHDC; ISSN: 0132-344X

PUBLISHER: DOCUMENT TYPE:

MAIK Nauka Journal LANGUAGE:

Russian

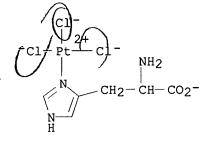
K2MCl4 (M = Pt, Pd) reacted with histidine (HL) in a 1:1 ratio in aq. AB soln. to give [MLC1(H2O)] (I), [Pt(HL)Cl2] (II), [Pt(H2L)Cl3] (III) and H3L[MC14]. The complexes were characterized by 1H, 14N and 195Pt NMR and IR spectra. I are mixts. of complexes of the same compn., contg. tautomeric forms of L-. For I (M = Pd) the mixt. was sepd. The mode of coordination of HL in II is analogous to that found in the Pd analog. In III H2L+ is coordinated through the tertiary N of the heterocycle.

ΙT 160208-88-8P

> RL: SPN (Synthetic preparation); PREP (Preparation) (prepn. of)

RN 160208-88-8 CAPLUS

CN Platinate(2-), trichloro(L-histidinato-N3)-, dihydrogen, (SP-4-2)- (9CI) (CA INDEX NAME)



H+

ANSWER 10 OF 26 CAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

1995:21466 CAPLUS

DOCUMENT NUMBER:

122:92534

TITLE:

Photochemistry of PtCl62- complex and reactions of intermediate Pt(III) complexes with bioorganic ligands

AUTHOR(S):

Plyusnin, V. F.; Grivin, V. P.

CORPORATE SOURCE:

Inst. Khim. Kinet. Goren., Novoskbirsk, 630090, Russia

SOURCE:

Khimiya Vysokikh Energii (1994), 28(3), 252-6

CODEN: KHVKAO; ISSN: 0023-1193

DOCUMENT TYPE:

Journal Russian

LANGUAGE:

- Spectral and kinetic characteristics were detd. of the transient Pt(III) complexes produced in laser photolysis of PtCl62- in alc. solns. Unstability of the coordination sphere of the Pt(III) ion allows to utilize photochem. of PtCl62- for a fast synthesis of the new coordination compds. of Pt(II). Bioorg. ligands (L) react fast with transient PtCl52to form PtCl52-...L. In many cases the ligands remained in the coordination sphere during Pt(III) .fwdarw. Pt(II) transformation, what led to photochem. synthesis of Pt(II) compds.
- IT 128507-00-6P

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (photochem. of PtCl62- complex and reactions of intermediate Pt(III) complexes with creatinine)

RN 128507-00-6 CAPLUS

CN Platinate(2-), (2-amino-1,5-dihydro-1-methyl-4H-imidazol-4-one-N3)pentachloro-, (OC-6-21)- (9CI) (CA INDEX NAME)

ANSWER 11 OF 26 CAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

1993:182038 CAPLUS

DOCUMENT NUMBER:

118:182038

TITLE:

Dehydrohalogenation and dimeric complex formation

proceeded at the solid-phase heating of

trans-[Pt(3-methyl-4-acetyl-5-aminopyrazole)(dimethyl

sulfoxide)Cl2]. X-ray structure of the di-.mu.-N1, N2-(3-methyl-4-acetyl-5-

aminopyrazolate) bis (dimethyl sulfoxide)dichlorodiplatinum(II)

AUTHOR(S):

Kukushkin, V. Yu.; Aleksandrova, E. A.; Leovac, Vukadin, M.; Iveges, Erika Z.; Belsky, Vitalii K.;

Konovalov, Vadim E.

CORPORATE SOURCE:

Dep. Chem., St. Petersburg State Univ., Stary

Petergof, 198904, Russia

SOURCE:

Polyhedron (1992), 11(20), 2691-6 CODEN: PLYHDE; ISSN: 0277-5387

DOCUMENT TYPE:

Journal

LANGUAGE: English

The bridge-splitting reaction of [Pt(Me2SO)(.mu.-Cl)Cl]2 with 3-methyl-4-acetyl-5-aminopyrazole (R3PzH) in acetone gives trans-[Pt(Me2SO)(R3PzH)Cl2] (I). On heating in the solid phase I undergoes thermal conversion to give [Pt2(R3Pz)2(Me2SO)2Cl2] (II) and the abstraction of HCl. The structure of II was detd. by x-ray diffraction anal. The complex exhibits a heat-to-tail dimeric structure with 2 bridged pyrazolate anions. The dimerization reaction represents an unusual type of solid-phase conversion of Pt complexes, which can be considered as a variant of the Anderson rearrangement. I and II were also characterized by elemental anal., IR spectral and TGA (I only) methods.

ΙT 147020-61-9P

> RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(prepn. and solid-state dimerization of)

RN 147020-61-9 CAPLUS

CN Platinum, [1-(3-amino-5-methyl-1H-pyrazol-4-yl)ethanone]dichloro[sulfinylb is[methane]-S]-, (SP-4-1)- (9CI) (CA INDEX NAME)

1 ANSWER 12 OF 26 CAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER: 1990:488018 CAPLUS

DOCUMENT NUMBER: 113:88018

TITLE: Pulsed laser photolysis of the hexachloroplatinate

(PtCl62-)-creatinine system in methanol

AUTHOR(S): Grivin, V. P.; Plyusnin, V. F.; Khmelinski, I. V.;

Bazhin, N. M.; Mitewa, M.; Bontchev, P. R.

CORPORATE SOURCE: Inst. Chem. Kinet. Combust., Novosibirsk, 630 090,

USSR

SOURCE: Journal of Photochemistry and Photobiology, A:

Chemistry (1990), 51(3), 371-7 CODEN: JPPCEJ; ISSN: 1010-6030

DOCUMENT TYPE: Journal

LANGUAGE: English

AB The pulsed laser photolysis (excimer laser, XeCl; 308 nm) of the PtCl62--creatinine-methanol system was studied. The formation of an intermediate PtIII species (PtCl52--cr, where cr = creatinine) was demonstrated and its decay kinetics were examd. Some kinetic and thermodn. data of the photoinduced reaction were detd. The photolysis of the same system using stationary irradn. was also investigated allowing the end product of the reaction to be detd.

IT 128507-00-6P

RL: FORM (Formation, nonpreparative); PREP (Preparation) (formation of, in photolysis of hexachloroplatinate-creatinine methanol system)

RN 128507-00-6 CAPLUS

CN Platinate(2-), (2-amino-1,5-dihydro-1-methyl-4H-imidazol-4-one-N3)pentachloro-, (OC-6-21)- (9CI) (CA INDEX NAME)

L21 ANSWER 13 OF 26 CAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER: 1989:184755 CAPLUS

DOCUMENT NUMBER: 110:184755

TITLE:

Dynamic and static trans effect of acetonitrile in

platinum(II) complexes

AUTHOR(S):

Krol, I. A.; Kukushkin, V. Yu.; Starikova, Z. A.;

Tkachuk, V. M.; Zhadanov, B. V.

CORPORATE SOURCE:

USSR

SOURCE:

Zhurnal Obshchei Khimii (1988), 58(11), 2625-6

CODEN: ZOKHA4; ISSN: 0044-460X

DOCUMENT TYPE:

Journal

LANGUAGE:

Russian

Et4N[Pt(MeCN)Cl3] (I) reacted with pyridine in aq. soln. to give trans-Pt(MeCN)pyCl2 (II) whereas I reacted with other amines to give cisand trans-Pt(MeCN)LC12 (III; L = benzimidazole, 2,5-dimethylpyrazole, PhNH2, p-anisidine). In Me2CO or MeCN trans-II is in equil. with cis-II. The formation of trans-III and the isomerization of trans-II confirm the high dynamic trans effect of MeCN in Pt(II) complexes. I is monoclinic, space group P21/c, with a 7.335(3) b 17.352(3), c 12.942(3) .ANG., .beta. 101.92(4).degree., dc = 1.984(3) g cm-3, Z = 4. The Pt-Cl bond for the Cl trans to MeCN is 2.262 .ANG. whereas the Pt-Cl bonds for cis Cl's are 2.293 and 2.301 .ANG., which indicate the high static trans effect of the Cl in comparison to MeCN. cis-Pt(MeCN)2Cl2 reacted with MCl to give M[Pt(MeCN)C13] (M = PPh4+, Ph3PCH2Ph+, AsPh4+, Et4N+, Bu4N+). Ph3PCH2Ph[Pt(MeCN)Cl3] was also obtained by dissoln. of (Ph3PCH2Ph)2[Pt2(.mu.-Cl)2Cl4] in MeCN or by the reaction of (Ph3PCH2Ph)2[Pt(NO2)Cl3] with Ti2(SO4)3 in MeCN-H2O. ΙT 119341-13-8P, cis-(Acetonitrile)dichloro(2,5-

dimethylpyrazole) platinum 119433-34-0P, trans-(Acetonitrile) dichloro (2,5-dimethylpyrazole) platinum RL: SPN (Synthetic preparation); PREP (Preparation) (prepn. of)

RN 119341-13-8 CAPLUS

CN Platinum, (acetonitrile)dichloro(1,3-dimethyl-1H-pyrazole-N2)-, (SP-4-3)-(CA INDEX NAME)

$$\begin{array}{c|c}
C1 - & \\
 & 2+ \\
 & C1 - Pt - N = C - Me
\end{array}$$
Me

N

Me

Me

RN 119433-34-0 CAPLUS

CN Platinum, (acetonitrile)dichloro(1,3-dimethyl-1H-pyrazole-N2)-, (SP-4-1)-(9CI) (CA INDEX NAME)

$$\begin{array}{c|c}
C1^{-} \\
\downarrow 2^{+} \\
\hline
-C1^{-} Pt \longrightarrow N \longrightarrow C^{-} Me
\end{array}$$
Me

N

Me

ANSWER 14 OF 26 CAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

1988:215259 CAPLUS

DOCUMENT NUMBER:

108:215259

TITLE:

Synthesis and characterization of new platinum(II) complexes containing thiazole and imidazole donors Muir, Mariel M.; Cadiz, Mayra E.; Baez, Adriana

AUTHOR(S): CORPORATE SOURCE:

Dep. Chem., Univ. Puerto Rico, Rio Piedras, 00932, P.

SOURCE:

Inorganica Chimica Acta (1988), 151(3), 209-13

CODEN: ICHAA3; ISSN: 0020-1693

DOCUMENT TYPE:

Journal English

LANGUAGE:

cis-Pt(NH3)LCl2 (L = thiazole, 2-bromothiazole, benzothiazole, 2,1,3-benzothiadiazole, 1,2,3-benzothiadiazole, imidazole,

1-methylimidazole) were prepd. The complexes were characterized by IR and UV-visible spectroscopy, 1H NMR and elemental analyses. The thiazoles and benzothiazoles were coordinated through the N heteroatom. Both the benzothiadiazoles were coordinated through S. Several of the complexes showed significant cytotoxic activity.

IT 114487-38-6P 114487-39-7P

RN 114487-38-6 CAPLUS

CN Platinum, amminedichloro(1-methyl-1H-imidazole-.kappa.N3)-, (SP-4-3)- (9CI) (CA INDEX NAME)

RN 114487-39-7 CAPLUS

CN Platinum, amminedichloro(1H-imidazole-N3)-, (SP-4-3)- (9CI) (CA INDEX NAME)

21 ANSWER 15 OF 26 CAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

1987:529989 CAPLUS 107:129989

DOCUMENT NUMBER: TITLE:

Assessment of DNA binding of platinum-radiosensitizer

complexes by inhibition of restriction enzymes

AUTHOR(S):

Skov, Kirsten A.; Adomat, Hans; Konway, Desmond C.;

Farrell, Nicholas P.

CORPORATE SOURCE:

Med. Biophys. Unit, British Columbia Cancer Res.

Cent., Vancouver, BC, V5Z 1L3, Can.

SOURCE:

Chemico-Biological Interactions (1987), 62(2), 117-29

CODEN: CBINA8; ISSN: 0009-2797

DOCUMENT TYPE: Journal LANGUAGE: English

A simple and rapid method has been used to compare the binding of Pt complexes to DNA, in a relatively qual. manner. A compd. bound at or near the restriction site inhibits enzymic cleavage of DNA; inhibition of BamHI and EcoRI activities by complexes was assessed in this study using linearized pSV2-gpt plasmid. The particular interest was in DNA binding by complexes of Pt with known org. radiosensitizers (RS), to det. whether the Pt was able to target the RS to the DNA. Although the PT-RS complexes investigated themselves have moderate radiosensitizing ability (like the inorg. complexes, cis- or trans-DDP), none of the Pt-RS inhibit to the same extent as cis- or trans-DDP. However, there appears to be some correlation between enhanced radiosensitization by Pt-RS over Pt(RS)2, with the degree of Pt binding (as assessed by the assay). The results using isolated DNA suggest that not all complexes bind well (e.g., Pt with 2 RS ligands), but that in certain cases (e.g., Pt with only 1 RS), it is possible to target the drug to the DNA. An ammine or amine ligand may be required to target a radiosensitizer to DNA using Pt.

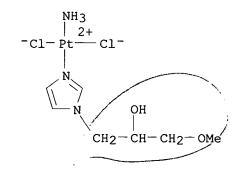
ΙT 110321-21-6

RL: BIOL (Biological study)

(DNA binding of, restriction enzymes inhibition in assessment of, radiosensitization in relation to)

110321-21-6 CAPLUS RN

CN Platinum, amminedichloro[.alpha.-(methoxymethyl)-1H-imidazole-1-ethanol-N3]-, (SP-4-3)- (9CI) (CA INDEX NAME)



ANSWER 16 OF 26 CAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

1985:497782 CAPLUS

DOCUMENT NUMBER:

103:97782

TITLE:

Platinum(II) complexes of nitroimidazoles: synthesis, characterization, and x-ray crystal structures of cis-dichlorobis[1-(2'-hydroxyethyl)-2-hydroxymethyl-5nitroimidazole]platinum(II) and trans-dichlorobis[1-

(2'-hydroxy-3'-methoxypropyl)-2-

nitroimidazole]platinum(II)

AUTHOR(S):

Bales, John R.; Mazid, Muhammed A.; Sadler, Peter J.; Aggarwal, Aneel; Kuroda, Reiko; Neidle, Stephen;

Gilmour, David W.; Peart, Barry J.; Ramsden,

Christopher A.

CORPORATE SOURCE:

SOURCE:

Dep. Chem., Birkbeck Coll., London, WC1E 7HX, UK Journal of the Chemical Society, Dalton Transactions: Inorganic Chemistry (1972-1999) (1985), (4), 795-802

CODEN: JCDTBI; ISSN: 0300-9246

DOCUMENT TYPE:

Journal English

LANGUAGE:

Twenty complexes PtL2X2 (I; L = substituted 5-nitroimidazole; X = Cl, Br, iodo; X2 = ethylmalonate, cyclobutane-1,1-dicarboxylate) were prepd. and characterized. I have a cis stereochem., as exemplified by the x-ray crystal-structure detn. of I [L = 1-(2'-hydroxyethyl)-2-(hydroxymethyl)-5nitroimidazole, X = Cl; II] using the heavy-atom method, with full-matrix least-squares refinement to R = 0.038 for 1102 obsd. reflections. Crystals of II are orthorhombic, space group Pcan, with a 8.643(1), b 24.052(3), c 9.119(1) .ANG., Z = 4, and d.(calcd.) = 2.243 g/cm3. Several analogous complexes of 2-nitro- and 4-nitroimidazoles were prepd. The geometry of the latter complexes was not detd., but the 2-nitroimidazoles form the thermodynamically favored trans complexes, rather than the kinetically favored cis complexes. This was verified by an x-ray crystal-structure detn. of trans-Pt(L1)2X2 [III; L1 = 1-(2'-hydroxy-3'methoxypropyl)-2-nitroimidazole, X = Cl]. Crystals of III are monoclinic, space group P21/a, with a 8.134(1), b 13.014(1), c 11.323(2) .ANG., .beta. 91.469(9).degree., and d.(calcd.) = 1.85 g/cm3; the structure was refined to R = 0.054 for 1706 obsd. reflections. There is an unusual loss of planarity in III between the NO2 group and the imidazole ring, giving a dihedral angle of 45.6.degree.. Coordination of the nitroimidazole ligand to Pt(II) lowered the wavelength of the .pi.-.pi.* electronic absorption band, and reduced the polarog. redn. potential by .apprx.0.15-0.2 V.

IT 97697-89-7P

RN 97697-89-7 CAPLUS

CN Platinate(1-), trichloro[.alpha.-(methoxymethyl)-2-nitro-1H-imidazole-1-ethanol-N3]-, potassium, (SP-4-2)- (9CI) (CA INDEX NAME)

 $m{2}$ 1 ANSWER 17 OF 26 CAPLUS COPYRIGHT 2004 ACS on STN.

ACCESSION NUMBER: 1986:80839 CAPLUS

DOCUMENT NUMBER: 104:80839

DOCUMENT NUMBER: 104:00039

TITLE: Synthesis and antitumor activity of platinum complexes

of protonated diamines

AUTHOR(S): Doran, Sheryl L.; Khokhar, Abdul R.; Hacker, Miles P.

CORPORATE SOURCE: Tumor Inst., M. D. Anderson Hosp., Houston, TX, 77025,

USA

SOURCE: Inorganica Chimica Acta (1985), 108(2), 113-15

CODEN: ICHAA3; ISSN: 0020-1693

DOCUMENT TYPE: Journal LANGUAGE: English

AB cis-[Pt(HQ)(L)Cl2], (Q = 3-aminoquinuclidine, N-aminopiperidine, piperazine, N-methylpiperazine, 1,1,4-trimethylpiperazine, and N-methyl-1,4-diazabicyclo[2.2.2]octane and L = SCN-, NO2-, Br-, and F-) were prepd. from [Pt(HQ)Cl3]. The antitumor activities of the complexes were evaluated in vitro against L1210 murine leukemia cells, and ID50 values for the L-substituted complexes were compared to values of the parent complexes. In each case replacement of a chloride ion by SCN-, NO2-, Br-, or F-, either reduced or completely eliminated antitumor

activity. This effect is explained in terms of the trans-directing ability of the ligand L, compared to chloride. The NO2-substituted complex of 3-aminoquinuclidine was tested in vivo and found to exhibit little or no antitumor activity.

IT 100226-99-1P 100227-00-7P 100227-01-8P 100227-04-1P

RL: BAC (Biological activity or effector, except adverse); BSU (Biological study, unclassified); SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES (Uses)

(prepn. and antitumor activity of, trichloro analog in relation to)

RN 100226-99-1 CAPLUS

● H+

RN 100227-00-7 CAPLUS

CN Platinate(1-), dichloro(1-methylpiperazine-N4)(thiocyanato-S)-, hydrogen, (SP-4-3)- (9CI) (CA INDEX NAME)

H+

RN 100227-01-8 CAPLUS

CN Platinate(1-), dichloro(1-methylpiperazine-N4)(nitrito-N)-, hydrogen, (SP-4-3)- (9CI) (CA INDEX NAME)

H+

RN 100227-04-1 CAPLUS

CN Platinum, dichloro(nitrito-N)(1,1,4-trimethylpiperazinium-N4)-, (SP-4-3)- (9CI) (CA INDEX NAME)

LP1 ANSWER 18 OF 26 CAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

1982:181154 CAPLUS

DOCUMENT NUMBER:

96:181154

TITLE:

Anticarcinogenic organic platinum complexes

Sankyo Co., Ltd., Japan

PATENT ASSIGNEE(S): SOURCE:

Jpn. Kokai Tokkyo Koho, 3 pp.

CODEN: JKXXAF

DOCUMENT TYPE:

Patent

LANGUAGE:

Japanese

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE

JP 56158763 A2 19811207 JP 1980-61522 19800509
PRIORITY APPLN. INFO.: JP 1980-61522 19800509

AB Stirring 0.338 g AgNO3 with 0.3 g cis-([pt(NH3)2Cl2] in H2O overnight at room temp. gave cis-[Pt(NH3)2(H2O)2](NO3)2, which was mixed with 0.111 g 2-mercaptopyridine in H2O, made pH 7 with 2 N aq. NaOH, and stirred with ice cooling to give [PtNH3(H2O)2(C5H4NS)](NO3) (I, C5H4NS = 2-mercaptopyridine residue) (yield not given). Also prepd. were Pt(NH3)(H2O)(C5H4NS)2 and (Pt(NH3)(H2O)2(C3H5N2S)](NO3) (C3H5N2S = ethylenethiourea residue). I at 100 mg/kg i.p. increased the survival time of sarcoma 180-infested mice by 244% over controls.

IT 81579-69-3P

RL: BAC (Biological activity or effector, except adverse); BSU (Biological

```
study, unclassified); SPN (Synthetic preparation); THU (Therapeutic use);
BIOL (Biological study); PREP (Preparation); USES (Uses)
   (prepn. and antineoplastic activity of)
81579-69-3 CAPLUS
Platinum(1+), amminediaqua(2-imidazolidinethionato-N1)-, (SP-4-3)-,
nitrate (9CI) (CA INDEX NAME)
CM
     1
CRN
     81579-68-2
```

CMF

CCI CCS

C3 H12 N3 O2 Pt S

RN

CN

CM

CRN 14797-55-8 CMF N O3

ANSWER 19 OF 26 CAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER: 1981:166780 CAPLUS

DOCUMENT NUMBER:

94:166780

TITLE:

Synthesis and antitumor activity of platinum complexes

with cyclic thioamides

AUTHOR(S):

Fujieda, Sigeaki; Tabata, Esuzu; Hatano, Akiko; Osa,

Tetsuo

CORPORATE SOURCE:

Pharm. Inst., Tohoku Univ., Sendai, 980, Japan

SOURCE:

Heterocycles (1981), 15(2), 743-6 CODEN: HTCYAM; ISSN: 0385-5414

DOCUMENT TYPE:

Journal

LANGUAGE:

English

The complexes [Pt(NH3)(H2O)2L]NO3 and [Pt(NH3)(H2O)L2](HL =2-mercaptopyridine (I), 2-thiouracil (II), ethylenethiourea (III)) were prepd. and their activity against Sarcoma 180 in mice was tested. Both complexes of I showed high antitumor activity but only the complex contg. 1 ligand of III was effective. The complexes of II had little antitumor activity.

ΙT 77319-89-2P

> RL: BAC (Biological activity or effector, except adverse); BSU (Biological study, unclassified); SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES (Uses)

(prepn. and antitumor activity of)

RN 77319-89-2 CAPLUS

Platinum(1+), amminediaqua(2-imidazolidinethionato-S)-, (SP-4-2)-, nitrate CN (CA INDEX NAME)

CM 1

CRN 77319-88-1

CMF C3 H12 N3 O2 Pt S

CCI CCS

CM 2

CRN 14797-55-8 CMF N O3

0= N-0-

LX1 ANSWER 20 OF 26 CAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER: 1978:145351 CAPLUS

DOCUMENT NUMBER:

88:145351

TITLE: AUTHOR(S):

New piperazinium complex salts with antitumour action

: Doadrio, A.; Craciunescu, D.; Ghirvu, G.

CORPORATE SOURCE:

Dep. Inorg. Anal. Chem., Fac. Pharm., Madrid, Spain Anales de Quimica (1968-1979) (1977), 73(7-8), 1042-6

SOURCE: Anales de Quimica (1968-1979)
CODEN: ANQUBU; ISSN: 0365-4990

DOCUMENT TYPE:

Journal

LANGUAGE:

English

AB The salts [LH2][PtC16], [LH2][OsC16], and [LH2][CuC14], where L = piperazine and its derivs. such as 1-methylpiperazine, 1-piperazineethanol, 1,4-bis(2-hydroxyethyl)piperazine, 4-(3-aminopropyl)-1-piperazineethanol, were prepd. and characterized by IR, magnetic, and conductimetric data, and their antitumor activities are reported. The electronic charge is localized to a much greater extent on the ring N atoms than on any exocyclic amino group.

IT 65703-85-7P 66098-32-6P

RL: BAC (Biological activity or effector, except adverse); BSU (Biological study, unclassified); SPN (Synthetic preparation); BIOL (Biological study); PREP (Preparation)

(prepn., electronic charge localization and antitumor activity of)

RN 65703-85-7 CAPLUS

CN Platinate(2-), hexachloro-, (OC-6-11)-, dihydrogen, compd. with 4-(3-aminopropyl)-1-piperazineethanol (1:1) (9CI) (CA INDEX NAME)

CM 1

CRN 58110-73-9 CMF C9 H21 N3 O

CM 2

CRN 16941-12-1 CMF C16 Pt . 2 H CCI CCS

2 H+

RN 66098-32-6 CAPLUS
CN Platinate(2-), hexachloro-, (OC-6-11)-, dihydrogen, compd. with 1-piperazineethanamine (1:1) (9CI) (CA INDEX NAME)

CM 1

CRN 16941-12-1 CMF C16 Pt . 2 H CCI CCS

●2 H+

CM 2

CRN 140-31-8 CMF C6 H15 N3

LX1 ANSWER 21 OF 26 CAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER: 1978:499758 CAPLUS

DOCUMENT NUMBER: 89:99758

TITLE: Effects of the compounds (PtCl6)H2(3-piperazinyl-1-

ethylamine) and (OsCl6) H2 (aminopyrine) 2 on calcium and

sodium transport in isolated frog skin

AUTHOR(S): Anadon, A.; Craciunescu, D.; Doadrio, A.; Larranaga,

M. R. M.; Sanz, F.

CORPORATE SOURCE: Fac. Vet., Univ. Madrid, Madrid, Spain

SOURCE: Acta Pharmaceutica Fennica (1977), 86(4), 203-7

CODEN: APHFDO; ISSN: 0356-3456

DOCUMENT TYPE: Journal LANGUAGE: English

AB A study was conducted on the effects of the salt complexes (PtCl6) H2(3-piperazinyl-1-ethylamine) [66632-41-5] and

(OsCl6)H2(aminopyrine)2 [65230-73-1] on elec. potentials, Na transport and Ca diffusion in frog skin used as a model membrane in an attempt to understand their pharmacol. effects. The results indicate that the salt complex (OsCl6)H2(aminopyrine)2 decreases both the transport of Na and elec. potential, and inhibits the effect of CaCl2. The salt complex

(PtCl6)H2(3-piperazinyl-1-ethylamine) does not exhibit any effect. IT 66632-41-5

RL: BIOL (Biological study)

(calcium and sodium transport by cell membrane response to)

RN 66632-41-5 CAPLUS

CN Platinate(2-), hexachloro-, (OC-6-11)-, dihydrogen, compd. with

4-ethyl-2-piperazinamine (1:1) (9CI) (CA INDEX NAME)

CM 1

CRN 66632-40-4 CMF C6 H15 N3

CM 2

```
CRN 16941-12-1
CMF
   Cl6 Pt . 2 H
CCI
   CCS
```

H+ 2

1/21 ANSWER 22 OF 26 CAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER: 1966:44120 CAPLUS

DOCUMENT NUMBER: 64:44120 ORIGINAL REFERENCE NO.: 64:8277b-c

TITLE: Aminolysis of sucrose. VI. Reaction between sucrose

and aqueous .beta.-aminopropionitrile solution at

increased temperatures

AUTHOR(S): Jezo, I.; Luzak, I.

CORPORATE SOURCE: Slovenska Akad. Vied, Bratislava, Czech. Chemicke Zvesti (1965), 19(12), 900-7 SOURCE:

CODEN: CHZVAN; ISSN: 0366-6352

DOCUMENT TYPE:

Journal LANGUAGE: Czech

cf. CA 63, 665f. Reaction of sucrose with aq. .beta.-aminopropionitrile soln. at higher temps. gave a mixt. of compds.: 2-methylpyrazine (b751 132-4.degree., n20D 1.5056; picrate m. 132-3.degree.) 2,5-dimethylpyrazine (b751 152-4.degree., n20D 1.5040; picrate 155-6.degree.), 2-methyl-1,4(?)-dihydropyrazine (b10 45-8.degree., n20D 1.5000, d20 1.0042; picrate m. 172.degree.), 4(5)-methylimidazole, and 2-methyl-4-(.beta.-cyanoethyl)-1,4-dihydropyrazine, b0.01 103-5.degree., n20D 1.5016. At the same time with the formation of these compds. there is a dismutation of .beta.-aminopropionitrile with the formation of bis(2-cyanoethyl)-amine. The reaction mechanism is given.

IT 7415-10-3, 1-Piperazinepropionitrile, 2-methyl-,

hexachloroplatinate(IV)

(prepn. of)

7415-10-3 CAPLUS

CN 1-Piperazinepropionitrile, 2-methyl-, hexachloroplatinate(IV) (7CI, 8CI)

CM 1

RN

CRN 16941-12-1 Cl6 Pt . 2 H CMF

CCI CCS

2 H⁺

CM 2

CRN 6216-09-7 CMF C8 H15 N3

ANSWER 23 OF 26 CAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER: 1965:29873 CAPLUS

DOCUMENT NUMBER: 62:29873

ORIGINAL REFERENCE NO.: 62:5322f-g

TITLE: Aminolysis of sucrose. IV. Reaction of sucrose with

aqueous solution of ethylenediamine

AUTHOR(S): Jezo, Ivan; Luzak, Ivan

CORPORATE SOURCE: Slovak Acad. Sci., Bratislava, Czech. SOURCE:

Chemicke Zvesti (1964), 18(3), 186-98

CODEN: CHZVAN; ISSN: 0366-6352

DOCUMENT TYPE: Journal LANGUAGE: Slovak

cf. CA 61, 2039a. The aminolysis of sucrose was carried out with an aq.

soln. of ethylenediamine. The following compds. were isolated and

identified: 2-methyl-2-imidazoline, 2-ethyl-2-imidazoline,

2-hydroxy-6,7-dihydro-8-azaquinolizine, piperazino[a]-3-hydroxy-8azaquinolizidine and piperazino[a]-3-hydroxy 8-aza-.DELTA.8,9(9,10)dehydroquinolizidine. A reaction mechanism is proposed.

IT 2870-74-8, 2-Piperazineacetonitrile, hexachloroplatinate(IV)

(prepn. of)

RN 2870-74-8 CAPLUS

CN 2-Piperazineacetonitrile, hexachloroplatinate (8CI) (CA INDEX NAME)

CM 1

CRN 16941-12-1 CMF C16 Pt . 2 H

CCI CCS

2 H⁺

CM 2

CRN 2465-79-4 CMF C6 H11 N3

$$\begin{array}{c}
H \\
N \\
CH_2-CN \\
H
\end{array}$$

121 ANSWER 24 OF 26 CAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

1962:469229 CAPLUS

DOCUMENT NUMBER:

57:69229

ORIGINAL REFERENCE NO.:

57:13756a-h

TITLE:

Synthesis of potential anticancer agents. XXX.

(1-Aziridinyl)purines

AUTHOR(S):

Montgomery, J. A.; Hewson, K.; Temple, C., Jr.

CORPORATE SOURCE:

Southern Res. Inst., Birmingham, AL

SOURCE:

Journal of Medicinal & Pharmaceutical Chemistry

(1962), 5, 15-24

CODEN: JMPCAS; ISSN: 0095-9065

DOCUMENT TYPE:

Journal Unavailable

LANGUAGE:
OTHER SOURCE(S):

CASREACT 57:69229

cf. CA 56, 1445a. Eight (1-aziridinyl) purines were prepd. by nucleophilic displacement of the Cl in the corresponding chloropurines. The compds. were screened against Ca 755 and Walker 256, and the 6-(1aziridinyl) purines show moderate activity against the Ca 755, and in 1 case good activity against Walker 256. 6-(1-Aziridinyl)-9-benzyl-9H-purine (I) (100 mg.) in 10 ml. EtOH was hydrogenated at room temp./atm. pressure with 50 mg. 5% Pd-C catalyst 18 hrs., the catalyst filtered off, the filtrate evapd. to dryness, the residue dissolved in 10 ml. 0.1N HCl, the soln. evapd. to dryness in vacuo, the residue triturated with 10 ml. Me2CO, the ppt. filtered off, and dried in vacuo to give 90 mg. 9-benzyl-N6-ethyl-adenine-HCl. I can also be reduced with metallic Na in liquid NH3 to give N6-ethyladenine, m. 227-8.degree. (decompn.). A mixt. of 5.7 g. 2,6-dichloropurine, 6.9 ml. benzyl chloride, 4.2 g. K2CO3, and 50 ml. Me2SO was stirred 18 hrs., filtered, the filtrate concd. to one-quarter vol. in vacuo, poured into ice-H2O, the oil allowed to solidify, filtered, washed with H2O, air dried, and recrystd. from 75% aq.

EtOH with Norit treatment to give 5.9 g. of a mixt. of the 7- and 9-isomers of benzyl-2,6-dichloro-9H-purine, m. 122-30.degree.. from MeOH gave 1.06 g. pure 9-isomer, m. 148.degree.. The mother liquors from the recrystn. gave 30% yield of the 7-isomer. contaminated with the 9-isomer. Prepd. similarly was 2-amino-9-benzyl-6-chloro-9H-purine, and 2-amino-7-benzyl-6-chloro-7H-purine. 9-Benzyl-2,6-di-chloropurine (278 mg.) in 10 ml. 0.2N naOH was refluxed with stirring to effect soln., the hot mixt. filtered, the filtrate acidified with glacial HOAc, the ppt. filtered off, washed with dil. HOAc, and dried in vacuo to give 233 mg. 9-benzyl-2-chlorohypoxanthine (II), m. 245.degree.. II (100 mg.) in 15 ml. abs. EtOH was placed in a glass-lined Parr bomb, satd. with dry NH3 at 5.degree., the soln. heated 6 hrs. at 150.degree., the bomb chilled, opened, the soln. concd. to 10 ml. in a stream of dry n, the solid filtered off, washed with EtOH, and dried in vacuo to give 53 mg. 9-benzylguanine (III), m. 303.degree.. III may also be prepd. from 2-amino-9-benzyl-6-chloropurine by reaction with HCl. II (0.29 millimole) and 0.2 ml. 25% aq. dimethylamine were dissolved in 10 ml. dioxane, the mixt. heated 20 hrs. at 95-100.degree., with an addn. of another quantity of Me2NH at the end of the first 2 hrs., the mixt. evapd. to dryness in vacuo, the residue triturated with Et2O, and the residue recrystd. from EtOH to give 57 mg. 9-benzylN2,N2-dimethylquanine, m. 282.degree.. 7-Benzyl-2,6-dichloropurine (contaminated with the 9-isomer) (1.11 q.) in 25 ml. n NaOH was heated 1.5 hrs. at 115.degree., cooled to room temp., kept overnight in the refrigerator, the solid filtered off, dissolved in H2O, acidified with glacial HOAc, the solid filtered off, dried in vacuo, and recrystd. from abs. EtOH to give 60 mg. 7-benzyl-2chlorohypoxanthine, m. 285.degree. 6,8-Dichloropurine (1.06 g.), 0.69 g. anhyd. K2CO3, and 1.4 ml. PhCH2Cl in 10 ml. HCONMe2 was heated 1 hr. at 70-85.degree. with stirring, concd. to 0.5 vol. in vacuo, poured into ice-H2O, the oil extd. with Et2O, the ext. dild. with EtOH, the solid filtered off, washed with Et2O, dried, extd. with Skellysolve C, the ext. evapd. to an oil, dissolved in 1:1 ag. EtOH, the solid filtered off, washed with H2O, dried in vacuo over P2O5 at room temp., and recrystd. from aq. EtOH to give 203 mg. 9-benzyl-6,8-dichoro-9H-purine, m. 92.degree.. Prepd. by similar methods was 9-benzyl-6-dimethylamino-9Hpurine, m. 117.degree.. 9-Benzyl-6-dimethylamino-8-chloropurine (307 mg.), 43 mg. MgO, and 34 mg. 5% Pd-C in 30 ml. abs. EtOH was hydrogenated at atm. pressure 6 hrs., the catalyst filtered off, the filtrate evapd. to dryness, and the residue triturated with EtOH to give 135 mg. 9-benzyl-6-dimethylaminopurine, m. 128.degree..

IT 94071-79-1, Piperazine, 1-(2-aminoethyl)-4-ethyl-, hexachloroplatinate(IV)

(prepn. of)

RN 94071-79-1 CAPLUS

Piperazine, 1-(2-aminoethyl)-4-ethyl-, hexachloroplatinate(IV) (7CI) (CA INDEX NAME)

CM 1

CN

CRN 16941-12-1 CMF Cl6 Pt . 2 H CCI CCS

2 H⁺

CM 2

CRN 4489-46-7 CMF C8 H19 N3

ANSWER 25 OF 26 CAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

1962:469228 CAPLUS 57:69228

DOCUMENT NUMBER:

57:13755f-i,13756a

ORIGINAL REFERENCE NO.: TITLE:

Condensation reaction of amino alcohols with imides.

IV. Condensation products of .beta.-

(cycloamino) ethanols with succinimide and with

phthalimide-formation of .beta.-

(cycloamino) ethylamines

AUTHOR(S):

Nakajima, Kazuo

CORPORATE SOURCE:

SOURCE:

Teikoku Chem. Ind., Mibukayogosho, Kyoto Nippon Kagaku Zasshi (1960), 81, 1129-32

CODEN: NPKZAZ; ISSN: 0369-5387

DOCUMENT TYPE:

LANGUAGE:

Unavailable

Journal

cf. CA 57, 8545h. Seven aminoalcs. (RNCH2CH2OH) (I), where RN = (CH2)4N (a), (CH2)5N (b), .beta.-tetrahydroquinoline (c), morpholino (d), 4-ethylpiperazino (e), 4-butylpiperazino (f), and 4-phenylpiperazino (g). I and phthalimide heated at 200.degree. while removing the H2O formed yielded RNCH2CH2N(CO)2C6H4-o (II) (RN, m.p., b10, m.p. of picrate, and decompn. point of Pt salt given): a, 109.degree., 220.degree., 218.degree., 212.degree.; b, 90.degree., 206.degree., 215.degree., 215.degree.; c, 137.degree., 273.degree., 144.degree., 204.degree.; d,

135.degree., 216.degree., 232.degree., 146.degree.; e, 115.degree.,

227.degree., 245.degree. (decompn.), 265.degree.; f, 75.degree.,

252.degree., 260.degree. (decompn.), 271.degree.; g, 158.degree., 279.degree., 218.degree. (decompn.), 262.degree.. Similarly, I and

```
succinimide yielded RNCH2CH2N(COCH2)2 (III) (RN, m.p., b10, m.p. of
picrate, and decompn. point of Pt salt given): a, 55.degree., 170.degree.,
158.degree., 212.degree.; b, -, 170.degree., 190.degree., 230.degree.; c,
115.degree., 210.degree., 180.degree., -; d, 75.degree., 170.degree.,
181.degree., 175.degree.; e, -, 199.degree., 245.degree., 245.degree.; f,
65.degree., 223.degree., 225.degree., 255.degree.; g, 141.degree.,
272.degree., 175.degree., 236.degree.. II and III refluxed with 20% HCl
3-5 hrs., phthalic acid or succinic acid filtered off, and the filtrate
concd. yielded RNCH2CH2NH2.HCl (IV.HCl). Free RNCH2CH2NH2 (IV) was
obtained by a long extn. of neutralized IV.HCl with ether.
b760, m.p. of picrate, and decompn. point of Pt salt given) a, 33.degree.,
165-6.degree., 210.degree. (decompn.), 207.degree.; b, 50.degree.,
185.degree., .apprx.200 (decompn.), 218.degree.; c, 157.degree.,
290.degree., -, -; d, 81.degree., 204.degree., 177.degree. (decompn.),
237.degree.; e, 88.degree., 223-4.degree., 230.degree., 270.degree.; f,
120.degree., 255.degree., 116.degree., 252.degree.; g, (m. 80.degree.),
191.degree., 326.degree., -. 210-40.degree..
94071-79-1, Piperazine, 1-(2-aminoethyl)-4-ethyl-,
hexachloroplatinate(IV) 94387-55-0, Piperazine,
1-(2-aminoethyl)-4-butyl-, hexachloroplatinate(IV)
   (prepn. of)
94071-79-1 CAPLUS
Piperazine, 1-(2-aminoethyl)-4-ethyl-, hexachloroplatinate(IV) (7CI)
INDEX NAME)
CM
     1
     16941-12-1
CRN
CMF
     C16 Pt . 2 H
CCI
    CCS
```

RN

CN

2 H+

CM 2

CRN 4489-46-7 CMF C8 H19 N3

RN 94387-55-0 CAPLUS

CN Piperazine, 1-(2-aminoethyl)-4-butyl-, hexachloroplatinate(IV) (7CI) (CA INDEX NAME)

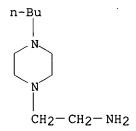
CM 1

CRN 16941-12-1 CMF C16 Pt . 2 H CCI CCS

2 H⁺

CM 2

CRN 4489-55-8 CMF C10 H23 N3



L21 ANSWER 26 OF 26 USPATFULL on STN

ACCESSION NUMBER:

2003:40793 USPATFULL

TITLE:

Process for preparing amine platinum complexes

INVENTOR(S):

Wong, Ernest S. Y., Langley, CANADA

Gianomenico, Christen M., Blaine, WA, United States

PATENT ASSIGNEE(S):

Anormed, Inc., Langley B.C., CANADA (non-U.S.

corporation)

NUMBER KIND DATE _____ US 6518428 B1 20030211 PATENT INFORMATION: 20000411 (9) APPLICATION INFO.: US 2000-547074

> DATE NUMBER ______

PRIORITY INFORMATION:

US 1999-128939P 19990413 (60)

DOCUMENT TYPE:

Utility

FILE SEGMENT:

GRANTED

PRIMARY EXAMINER: LEGAL REPRESENTATIVE: Huang, Evelyn Mei

NUMBER OF CLAIMS:

Morrison & Foerster LLP

EXEMPLARY CLAIM:

33 1

NUMBER OF DRAWINGS:

9 Drawing Figure(s); 5 Drawing Page(s)

LINE COUNT:

1148

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

The present invention relates to the area of platinum drugs. In particular, it relates to an improved process for preparing platinum complexes having the general formula (Ia) or (Ib): as defined of herein, ##STR1##

comprising:

- 1a) a first step, wherein [PtA.sub.4].sup.2- is reacted with L under appropriate conditions in a first solvent to form [PtA.sub.3(L)].sup.-;
- 1b) a second step, wherein [PtA.sub.3(L)].sup.- is reacted with L' under appropriate conditions in a second solvent to form cis-[PtA.sub.2(L')(L)];
- 1c) in the case where Y is halogen or hydroxy a third step, wherein cis-[PtA.sub.2(L')(L)] is reacted with H.sub.20.sub.2, Y.sub.2 or halogen containing oxidant to form c,t,c-[PtA.sub.2Y.sub.2(L')(L)]; in the case where Y is carboxylate, carbamate or carbonate ester a fourth step, wherein an intermediate, where Y is hydroxy formed in step 1c), is functionalized with an appropriate acylating agent; and
- 1d) in the case where A is not a halide or is different from the original halide, additional step(s) in which the original halide A of an intermediate formed in step 1a or 1b 1c or 1d is converted to a different halide or a new leaving group(s) A such as mono-dentate hydroxy, alkoxy, carboxylate or bi-dentate carboxylate, phosphonocarboxylate, diphosphonate, or sulphate.

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

301299-27-4P 301299-34-3P

(improved prepn. of antitumor agent)

RN 301299-27-4 USPATFULL

Platinum, amminedichloro(2,5-dimethylpyrazine-.kappa.N1)-, (SP-4-3)- (9CI) CN (CA INDEX NAME)

RN 301299-34-3 USPATFULL

CN Platinum, amminedichloro(2,3-dimethylpyrazine-.kappa.N1)dihydroxy-, (OC-6-43)- (9CI) (CA INDEX NAME)

IT 301299-38-7

(reactant for improved prepn. of platinum amine complexes as antitumor agents)

RN 301299-38-7 USPATFULL

CN Platinum, amminedichloro(2,3-dimethylpyrazine-.kappa.N1)-, (SP-4-3)- (9CI) (CA INDEX NAME)

FILE 'CAOLD' ENTERED AT 16:14:13 ON 07 JAN 2004
USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT.
PLEASE SEE "HELP USAGETERMS" FOR DETAILS.
COPYRIGHT (C) 2004 AMERICAN CHEMICAL SOCIETY (ACS)

FILE COVERS 1907-1966 FILE LAST UPDATED: 01 May 1997 (19970501/UP)

This file contains CAS Registry Numbers for easy and accurate substance identification. Title keywords, authors, patent assignees, and patent information, e.g., patent numbers, are now searchable from 1907-1966. TIFF images of CA abstracts printed between 1907-1966 are available in the PAGE display formats.

This file supports REG1stRY for direct browsing and searching of all substance data from the REGISTRY file. Enter HELP FIRST for more information.

```
L10
         165325 SEA FILE=REGISTRY ABB=ON
                                          16.239/RID
L11
         371280 SEA FILE=REGISTRY ABB=ON
                                          16.165/RID
L12
         490595 SEA FILE=REGISTRY ABB=ON
                                         46.383/RID
L13
         567205 SEA FILE=REGISTRY ABB=ON
                                          16.195/RID
L15
            172 SEA FILE=REGISTRY ABB=ON
                                         PT/ELS AND ((3/N AND (L11 OR L12 OR
                L13)) OR (2/N AND L10))
             52 SEA FILE=REGISTRY ABB=ON L15 AND 1/NR
L16
L20
              4 SEA FILE=CAOLD ABB=ON L16
```

=> d iall hitstr 120 1-4

L20 ANSWER 1 OF 4 CAOLD COPYRIGHT 2004 ACS on STN

CA64:8277b CAOLD

ACCESSION NUMBER:

TITLE:

aminolysis of sucrose - (VI) reaction between sucrose and aq. .beta.-aminopropionitrile soln. at increased temps.,

(VII) reaction between sucrose and an aq. soln. of

dimethylamine at higher temps.

AUTHOR NAME:

INDEX TERM:

Jezo, Ivan; Luzak, I.

151-18-8 6216-07-5

6216-08-6

6216-09-7 6216-10-

6216-15-5 6372-03-8 7365-19-7 7415-09-0

7415-10-3 7437-30-1 91144-97-7

IT **7415-10-3**

RN 7415-10-3 CAOLD

CM 1

CRN 16941-12-1 CMF C16 Pt . 2 H CCI CCS

2 H+

CM 2

CRN 6216-09-7 CMF C8 H15 N3

L20 ANSWER 2 OF 4 CAOLD COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER: CA62:5322g CAOLD

TITLE:

prepn. and use of .beta.-cyanoethylated polyhydric alcs. in

gas chromatography

AUTHOR NAME:

Ratusky, Josef; Bastar, L.

INDEX TERM:

89-71-4	93-58-3	93-60-7	98-27-1	99-36-5
99-75 - 2	99-76-3	100-23-2	119-36-8	534-26-9
611-20-1	612-24-8	617-37-8	619-24-9	626-86-8
767-00-0	873-62-1	1070-34-4	1070-62-8	1071-46-1
1129-89-1	1459-93-4	1486-75-5	2305-59-1	2409-55-4
2459-05-4	2459-07-6	2459-09-8	2459-10-1	2459-24-7
2459-25-8	2465-78-3	2465-79-4	2465-80-7	2465-81-8
2465-82-9	2465-85-2	2465-86-3	2465-87-4	2465-88-5
2465-89-6	2465-91-0	2465-92-1	2465-93-2	2465-94-3
2484-57-3	2484-58-4	2484-59-5	2484-60-8	2484-61-9
2484-62-0	2484-64-2	2484-65-3	2484-66-4	2484-74-4
2671-74-1	2672-57-3	2672-58-4	2778-62-3	2779-94-4
2870-74-8	3990-03-2	19438-10-9	90418-99-8	
94033-42-8	94409-05-9	94409-06-0		

94033-42-8 94409-05-9 94409-06-0

ΙT 2870-74-8

RN2870-74-8 CAOLD

CN2-Piperazineacetonitrile, hexachloroplatinate (8CI) (CA INDEX NAME)

CM

CRN 16941-12-1 CMF Cl6 Pt . 2 H

CCI CCS

2 CM

CRN 2465-79-4 CMF C6 H11 N3

L20 ANSWER 3 OF 4 CAOLD COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER: CA57:13756a CAOLD

TITLE:

synthesis of potential anticancer agents - (XXX)

(1-aziridinyl) purines

AUTHOR NAME:

Montgomery, John A.; Hewson, K.; Temple, C., Jr.

INDEX TERM:

4489-46-7 4489-55-8 6311-27-9 6332-42-9 6336-42-1 7461-80-5 14937-72-5 21091-61-2 53665-01-3 56025-87-7 56025-88-8 79064-26-9 90559-83-4 91395-18-5 92193-48-1

92291-23-1 92291-24-2 92495-26-6 93256-54-3

94387-55-0 95194-86-8 94071-79-1

96952-06-6 97619-37-9

94071-79-1 94387-55-0 ΙT

RN94071-79-1 CAOLD

CN Piperazine, 1-(2-aminoethyl)-4-ethyl-, hexachloroplatinate(IV) (7CI) INDEX NAME)

CM 1

16941-12-1 CRN CMF Cl6 Pt . 2 H CCI CCS

2 H+

2 CM

CRN 4489-46-7 CMF C8 H19 N3

RN 94387-55-0 CAOLD

CM 1

CRN 16941-12-1 CMF C16 Pt . 2 H CCI CCS

●2 H+

CM 2

CRN 4489-55-8 CMF C10 H23 N3

L20 ANSWER 4 OF 4 CAOLD COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER: CA51:16490f CAOLD

TITLE:

triethylenediamine (1,4-diazabicyclo-[2,2,2]octane) and

hexaethylenetetramine - (III) interaction of

2,2',2'',-trichlorotriethylamine-HCL and dimethylamine

```
AUTHOR NAME:
INDEX TERM:
```

Mann, Frederick G.; Baker, F. C. 104-19-8 6952-20-1 283**-**74-9 3644-18-6 5450-74-8 7699-64-1 14870-72-5 14968-74-2 14968-80-0 19479-83-5 22746-07-2 24996-75-6 36220-12-9 41809-88-5 60013-06-1 60420-20-4 65420-17-9 77267-14-2 96819-72-6 98545-14-3 98545-15-4 98545-16-5 99178-84-4 101433-80-1 101738-70-9 101787-23-9 101882-89-7 101882-90-0 103033-84-7 103163-44-6 103327-67-9 103401-77-0 103402-09-1 103510-04-9 107773-52-4 108129-20-0 108131-10-8 109407-12-7 109499-70-9 110055-62-4 110357-87-4 112224-31-4 112745-27-4 112745-30-9 114063-46-6 114225-70-6 114791-37-6 114985-11-4 114985-13-6 114985-51-2 115307-78-3 116032-45-2 116597-68-3 116928-68-8 117891-78-8 **122238-94-2** 122337-98-8 122679-89-4 123935-46-6 132984-24-8 133133-15-0

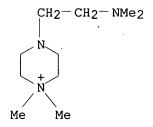
ΙT 122238-94-2

RN 122238-94-2 CAOLD

CN Bis[4-(2-dimethylaminoethyl)-1,1-dimethylpiperazinium] chloroplatinate(IV) (6CI) (CA INDEX NAME)

CM 1

CRN 122238-93-1 CMF C10 H24 N3



CM 2

CRN 16871-54-8 CMF Cl6 Pt CCI CCS

=> fil hom

FILE 'HOME' ENTERED AT 16:14:29 ON 07 JAN 2004

2,3:4,5-di-O-isopropylidene-6-O-trityl-p-galactose, $[\alpha]_3^6-28.6^\circ$ (c 1.5, CHCl₃), was prepd. I was isolated on preparative chromatography in CHCl₃ on a thin layer of Al₂O₂ as a sirup with R_f 0.64 and $[\alpha]_3^{26}-53.2^\circ$ (c 1, CHCl₃). Jan Micka

Sucrose esters. I. Chemical properties and EHL. J. Peris and J. M. Sune (Fac. Farm., Granada, Spain). Galenica Acta (Madrid) 17(3-4), 227-38(1964)(Span). Com. sucrose 6 and (or) 6' mono- and di-esters with fatty acids were tested for acid index, sapon. index, and acid index of hydrolyzed fatty acids, noting purity and EHL [= 20(1 no. 14 sapon. index/acid index of esterifying acid)]. Free acidity of the esters was relatively low, and EHL values fell between those of sorbitan esters (Spans) and poly(oxyethylene) derivs. (Tweens).

Keith S. Brown Aminolysis of sucrose. VI. Reaction between sucrose and aqueous β-aminopropionitrile solution at increased temperatures. I. Jezo and I. Luzak (Slovenska Akad. Vied, Bratislava, Czech.). Chem. Zvesti: 19(12), 900-7(1965)(Czech); cf. CA 63, 665f. Reaction of sucrose with aq. β-aminopropionitrile soln. at higher temps. gave a mixt. of compds.: 2-methylpyrazine (b₁₈; 132-4°, n₁°, 1.5056; picrate m. 132-3°) 2,5-dimethylpyrazine (b₁₈; 132-4°, n₂°, 1.5040; picrate 155-6°), 2-methyl-1,4(?)-dihydropyrazine (b₁₀, 45-8°, n₂°, 1.5000, d₂₀, 1.0042; picrate m. 172°), 4(5)-methylimidazole, and 2-methyl-4-(β-cyanoethyl)-1,4-dihydropyrazine, b_{0.01} 103-5°, n₂°, 1.5016. At the same time with the formation of these compds. there is a dismutation of β-aminopropionitrile with the formation of bis(2-cyanoethyl)-amine. The reaction mechanism is given. Jan Micka

amine. The reaction mechanism is given. Jan Micka c Aminolysis of sucrose. VII. Reaction between sucrose and an aqueous solution of dimethylamine at higher temperatures. I. Jezo and I. Luzak (Slovenska Akad. Vied, Bratislava, Czech.). Chem. Zvesti 19(12), 908-17(1965)(Czech); cf. preceding abstr. From the mixt., formed by the reaction of sucrose with an aq. soln. of Me₃NH at higher temps., the following compds. were isolated and identified: Me₃N, Me₃NEt, 2-dimethylamino-ethanol, 1,2-bis(dimethylamino)propane, 1,1,2-tris(dimethylamino)ethane, AcNMe₃ N,N-dimethylglycolamide, N,N,N',N' tetramethylglycinamide, and bis(dimethylamide) of cis(?)-tetrahydrofuran-2,5-dicarboxylic acid. A reaction mechanism d Jan Micka

is given.

Polysaccharides from Plantago major leaves. I. Analysis of monosaccharide composition of polysaccharide complex.

A. G. Gorin (Chem.-Pharm. Inst. Kharkov). Khim. Prirodn.

Soedin., Akad. Nauk Uz. SSR 1965(5), 297-302(Russ). A polysaccharide of the pectin type was isolated from the title—material. Dry leaves were extd. with boiling 80% EtOH and MesCO, and air-dried. The polysaccharide was extd. with boiling H₂O (1:15) for 2 hrs. and pptd. with 3 vols. of EtOH (yield 10%, ash 28%). The crude prepn. (10 g.) was dissolved in H₂O (100 cc.) and percolated through a cation-exchange resin KU-2 (H+) column (20 × 180 mm.). The demineralized soln. was poured in 4 vols. of EtOH and the pptd. polysaccharide dried by solvent exchange, yielding 4.5 g. of a prepn. [a]b +195.0° (c 0.43, 0.3N NaOH), galacturonic acid content by decarboxylation 68%, equiv. wt. 454, equiv. wt. after sapon. 263, methoxyl content 4.2%. A part of the polysaccharide was hydrolyzed with 1.0N H₂SO, for 8 hrs.; p-galacturonic acid, p-galactose, L-arabinose, and L-rhamnose were characterized as cryst. derivs. Small amts. of glucose, xylose, and 3 unknown compds., one of them being identified as a di-O-methylgalactose, were also found.

Aminocyclitols. IX. The facile synthesis of streptamine. f

Aminocyclitols. IX. The facile synthesis of streptamine. f Tetsuo Suami and Seiichiro Ogawa (Keio Univ., Tokyo). Bull. Chem. Soc. Japan 38(11), 2026(1965)(Eng); cf. CA 63, 8464f. Streptamine was prepd. via a new route. (±)-1,2:3,4-Di-O-isopropylidene-epi-inositol (I) (Angyal and Gilham, CA 52, 11759g) was converted to II, m. 147.5-8.5°, by treatment

with excess MeSO₂Cl in pyridine. If was heated on a water h bath 2 hrs. in 50% HOAc and the product acetylated to give (±)-5,8-di-O-mesyl-epi-inositol tetracetate (III), m. 166.5-8.5°. Treatment of III with NaN₃ in boiling aq. 2-methoxyethanol for 40 hrs. and subsequent acetylation afforded 28%

4,6-diazido-4,6-dideoxy-myo-inositol tetraacetate (IV), m. 147-9°. Catalytic hydrogenation of IV in EtOH and acetylation of the product gave hexaacetyl-4,6-myo-inosadiamine (V), m. 290-2°. V was selectively deacetylated to VI which was oxidized with Pt black in a stream of O for 24 hrs. at 40° to give VII. VII was immediately reduced with Na/Hg in a slightly acidic soln. and the redn. product acetylated to give 12.5% hexaacetylstreptamine (VIII), transition point 243-8°.

O. L. Marrs

Chemistry of 3,5-cyclohexadiene-1,2-diol. XIII. Synthesis of deoxyinosadiamines. Minoru Nakajima, Akira Hasegawa, and Takashi Kurokawa (Univ. Kyoto, Japan). Ann. Chem. 689, 229-34(1965)(Ger); cf. CA 62, 14796h. From 4 stereoisomeric N-acetyl-tri-O-acetylkonduramines (acetyl derivs. of trihydroxyaminocyclohexene) (CA 58, 9216g) were prepd. acetamido-tri-O-acetylinosamines by catalytic debromination of the resp. bromohydrins. These on oxidn. gave 4 stereoisomeric deoxyinosaminoses, 3 of which were converted into deoxyinosadiamines via their oximes. The assigned configurations were detd. on the basis of N.M.R. spectra. To 4 g. N-acetyl-tri-O-acetylkonduramine A 1 (I) (m. 157°) in 70 cc. H₂O and 70 cc. AcOH was added dropwise 70 cc. 3% H₂O-Br with ice cooling and the soln. let stand 2 hrs. at room temp. to give 3.6 g. II, m. 114-15°. Acetylation of II with Ac₂O-C₂H₃N gave 90%

pentaacetyl deriv., m. 212–13°, identical (mixed m.p. and ir spectrum) with authentic material. N-Acetyl-tri-O-acetylconduramine C 4 (III) (m. 165°) (1.8 g.) in 30 cc. H₂O and 30 cc. AcOH treated with 32 cc. 3% H₂O-Br gave 1.7 g. IV, m. 200–2° (EtOH). Acetylation of 41 mg. IV with Ac₂O-C₄H₄N gave 42 mg. pentaacetyl deriv., m. 180–1° (EtOH). Treatment of 6.3 g. N-acetyl-tri-O-acetylconduramine F 4 (V) (m. 148°) in 100 cc. H₂O and 100 cc. AcOH with 110 cc. 3% H₂O-Br gave 6.4 g. VI, 194° VI (41 g.) acetylated with Ac₂O-C₄H₄N gave 39 mg. pentaacetyl, deriv. m. 196° (EtOH). Treatment of 3.8 g. N-acetyl-tri-O-acetylconduramine B 1 (VII) (m. 158°)

43—CARBOHYDRATES

Study of kinetics of formaldehyde condensation into sugars with the aid of a high-frequency apparatus. I. L. Orestov a (Pharm. Inst., Pyatigorsk). Zh. Fiz. Khim. 38(11), 2664-6 (1964)(Russ); cf. Zarinskii and Koshkin, CA 53, 5b. The kinetics of CH₂O condensation into sugars in the presence of Ca-(OH), was studied at 40-60° by the change in the current i with time r in a high-frequency titration bridge-based app. (loc. cit.). The i vs. r plot consisted of 4 sections, a rising section during the 1st 20 min., a horizontal section for about 100 min., a section rapidly descending to a pointed min. at r in about 215 min., and a final rising section. The mechanism corresponding to each section was considered. Only at the min. was CH₂O absent. A min. before the min sugars were absent. They appeared in the presence of PtO₂ to the l-azide deriv. of I, b_{0.00} 62-5°, (a]²0 -46° (c 0.5, CHCl₂), which was quant. hydrogenated in the presence of PtO₂ to yield 1-amino-2,5-anhydro-1-deoxy-3,4-O-isopropylidene-p-ribitol (II); 2,4 dinitrobenzene deriv. m. 144.5-5.5° (EtOH), [a]p -63.6° (c 0.42, CHCl₂). Aq. HCl hydrolysis of II gave 1-amino-2,5-anhydro-1-deoxy-p-ribitol, m. 164-8°, [a]²0 -79° (c 0.442, H₂O).

Chromatographic purification of N-acetyl-N-p-glucopyranosyl-p-aminoazobenzene prepared by the aminolysis of N-acetyl-N-glucopyranosyl-p-aminoazobenzene prepared by the aminoazobenzene prepared by the aminoazobenzene prepared by the aminoazobenzene prepared by the aminoazobenzene prepared by the amino

5321

Synthesis of mychose (6-deoxy-2,3-di-O-methyl-p-allose). J. S. Brimacombe, M. Stacey, and L. C. N. Tucker (Univ. Birmingham, Engl.). J. Chem. Soc. 1964(Dec.), 5391-2(Eng). Mycinose, a sugar component of the antibiotic chalcomycin, has

been synthesized and its identity with 6-deoxy-2,3-di-O-methylp-allose (I) confirmed.

Partial acylation of 3-deoxy-D-xylo-hexose dithioacetals. Gerhard Rembarz and Thea Reinhard (Univ. Rostock, Ger.).

J. Prakt. Chem. 26(1-2), 79-82(1964)(Ger). Acylation of a 3-deoxy-D-xylo-hexose dialkyl dithioacetal (I) with a limited amt. of BzCl or p-toluenesulfonyl chloride gave a cryst. 6-benzoate (II) or a cryst. 6-p-toluenesulfonate (III). Dissolving 0.01 mole I in 50 cc. C₄H₃N, cooling to -15°, adding in 30 min. with stirring a soln. of 1.15 cc. BzCl in 5 cc. C₄H₃N, stirring 8 hrs. at d-10°, leaving at 0° 16 hrs. and at 20° 3 hrs., pouring into 200 cc. 10% AcOH previously cooled to 0°, recrystg., when crystals were formed, from CHCl₃ and pentane, dissolving sirups formed in CHCl₃, washing, and drying the soln. with Na₃SO₄, treating with active C, filtering through kieselguhr, evapg., and recrystg.—the residue from CHCl₃ and pentane gave the following II [alkyl, % yield, m.p., [a]\frac{3}{2}\cdot (C₄H₄N) given]: Me, 25, 125°, 34.7° (c 2.27); Pr, 38, 59°, 17.2° (c 1.47); Bu, 26, 66°, 28.4° (c 2.36). Proof of the structure of II (alkyl = Pr) was realized by shaking 12 hrs. 100 mg. of the compd. with 240 mg. Pb(OAc), in 60 cc. abs. C₄H₄, filtering through kieselguhr, evapg. the fil-

Chromatographic purification of N-acetyi-N-D-glucopyranosyl-p-aminoazobenzene prepared by the aminolysis of N-acetyl-N-[2,3,4,6-tetra-O-acetyl-D-glucopyranosyl]-p-aminoazobenzene. Janusz Sokolowski and Zofia Fialkiewicz (Wyzsza Szkola Pedagogiczna, Gdansk, Poland). Rozeniki Chem. 38(9), 1311-15 (1964)(Pol). A suspension of 2.485 g. N-acetyl-N-[2,3,4,6-tetra-O-acetyl-D-glucopyranosyl]-p-aminoazobenzene in 250 ml. anhyd. MeOH was treated with a soln. of 0.945 g. Me₂NH (I) in anhyd. MeOH was treated with a soln. of 0.945 g. Me₂NH (I) in a few ml. MeOH, stirred overnight at room temp., stripped at 15-20 mm. and repeatedly treated with MeOH and stripped to remove unreacted I and AcNMe₂. The crude N-acetyl-N-p-glucopyranosyl-p-aminoazobenzene (II) thus obtained was purified by chromatography to give 0.06 g. pure II, m. 115-20°, [a]²⁰⁰⁰₈₀₀₀ 48° (c 0.5, EtOH). A soln. of 1.64 g. II in 8 ml. C₄H₃N treated at room temp. with 2.8 g. BzCl, kept 24 hrs. at room temp., poured into 250 g. ice-water, and cooled several hrs. gave 0.9 g. N-acetyl-N-[2,3,4,6-tetra-O-benzoyl-p-glucopyranosyl]-p-aminoazobenzene, m. 169-71°, [a]²⁰⁰⁰₈₀₀₀ -82.5° (c 0.2, EtOH). Pure II (1.64 g.), 3.4 g. PhCHO, and 0.95 g. powd. anhyd. ZnClwas shaken 5 hrs. at room temp., then extd. successively several times with cold H₂O and petr. ether. The residue after evapn. was powd. and washed with aq. NaHCO₄ and H₂O to give 1.7° g. N-acetyl-N-[4,6-O-benzylidene-D-glucopyranosyl]-p-aminoazobenzene (III), m. 123-30°, [a]²⁰⁰⁰₈₀₀₀ (c 0.5, EtOH). To a cold soln. of 0.196 g. III in 3 ml. C₄H₅N, 3 ml. Ac₂O was added at 0°, the whole left 24 hrs. at room temp., then poured into 50 g. ine-water. to affect 0.21 g. N-acetyl-N-[4,6-D-benzylichen-p-glucopyranosyl]-p-amino-grounders. to affect 0.21 g. N-acetyl-N-[4,6-D-benzylichen-p-glucopyranosyl]-p-amino-grounders.

cold soln. of 0.196 g. III in 3 ml. C₆H₅N, 3 ml. Ac₇O was added at 0°, the whole left 24 hrs. at room temp., then poured into 50 g. ice-water, to afford 0.21 g. N-acetyl-N-[4,6-O-benzylidene-2,3-di-O-acetyl-D-glucopyranosyl]-ρ-aminoazobenzene, m. 137-40° (MeOH), [α]₈₀₀ - 147° (¢ 0.15, EtOH). W. Sobotka Novel synthetic approach to the ylidene derivatives of carbohydrates. E. Bergonzi, R. Bernetti, C. Boffi, V. Brocca, and E. A. Cleveland (Corn Prod. Co., Argo, III.). Staerke 16(12), 388-92(1964)(Eng). Cyclic acetals were prepd. from D-glucose (I) and Me α-D-glucopyranoside (II) in Me₃SO with BF₃.Et₂O catalyst. Under the same conditions, ketones did not react. II (0.2 mole), 0.6 mole EtCHO, and 2 ml. BF₃.Et₂O in 200 ml. Me₃SO was stirred 7 hrs. at 65° to give 2 g. Me 2,3-O-oxidodipropylidene-4,6-O-propylidene-α-D-glucopyranoside, m. 138-9°, Me₃SO was stirred 7 hrs. at 65° to give 2 g. Me 2,3-O-oxidodipropylidene-4,6-O-propylidene-α-D-glucopyranoside, m. 138-9°, [α]¹/₈ 72.2° (CHCl₃), and 35 g. Me 4,6-O-propylidene-α-D-glucopyranoside, m. 98-100°, [α]¹/₈ 118.2° (CHCl₃). Similarly was obtained Me 4,6-O-laurylidene-α-D-glucoside, m. 75-6° (MeOH), [α]¹/₈ 77.8° (EtOH), which with Ac₃O in C₃H₄N gave a diacetate, m. 45-6° (EtOH-H₁O), [α]¹/₈ 87.5° (EtOH). From I was obtained 4,6(?)-O-propylidene-D-glucose, m. 158-60° (EtOA-), [α]¹/₈ 4.3° (at equil., H₂O), and 4,6(?)-O-laurylidene-D-glucose, m. 136° (Me₂CO), [α]¹/₈ 36° (EtOH). 1,2-O-Isopropylidene-α-D-glucofuranose was obtained in 20% yield from 0.05 mole I, 0.1 mole Me₂CO, and 0.1 mole BF₃-Et₂O in 50 ml. dioxane at 62° for 1 hr. Similarly, 1,2-O-trichloroethylidene-D-glucofuranoses were prepd. in 80.8% yield (58.2% α and 22.6% β isomer).

Frederick W. Parrish Aminolysis of sucrose. IV. Reaction of sucrose with aqueous

Frederick W. Parrish Aminolysis of sucrose. IV. Reaction of sucrose with aqueous solution of ethylenediamine. Ivan Jezo and Ivan Luzak (Slovak Acad. Sci., Bratislava, Czech.). Chem. Zvesti 18(3), 186-98(1964)(Slo); cf. CA 61, 2039a. The aminolysis of sucrose was carried out with an aq. soln. of ethylenediamine. The following compds. were isolated and identified: 2-methyl-2-imidazoline, 2-hydroxy-6,7-dihydro-8-azaquinolizine, piperazino[a]-3-hydroxy-8-azaquinolizidine and piperazino[a]-3-hydroxy 8-aza-26.0(0.00)-dehydroquinolizidine. A reaction mechanism is proposed.

Preparation and use of 8-cvanoethylated polyhydric alcohological

mechanism is proposed.

Preparation and use of β-cyanoethylated polyhydric alcohols in gas chromatography. J. Ratusky and L. Bastar (Ceskoslov. Akad. Ved, Prague). Collection Czech. Chem. Commun. 29 (12), 3066-80(1964)(Ger). Addn. of CH₂:CHCN (I) to polyhydric alcs. was modified to avoid the undesired formation of polyacrylonitrile. The polyhydric alc. was heated with a catalytic amt. of NaOEt at 100° under simultaneous removal of EtOH. The resulting partial alkoxide was treated dropwise with I (at the reflux temp. of I) to give a low-melting mixt. of the starting polyhydric alc. with the partially β-cyanoethylated polyhydric alc. The temp. of the resulting homogeneous mixt. was then lowered and the remaining I was added. Thus, 24.2 g. finely powd. erythritol was stirred at 100° with 1.1 g. NaOEt, the EtOH being simultaneously distd., the residue treated dropwise in 5 min. at 80° with 15 ml. I and then in 45 min. at 60° with 37 ml. I, the whole kept overnight at room temp., decompd. with 100 ml. whole kept overnight at room temp., decompd. with 100 ml. H₂O, the suspension washed with CH₂Cl₂, the ppt. collected, and

Cholis on a

the filtrate extd. with CH₂Cl₁ (the CH₂Cl₂ exts. were evapd. and the residues β -cyanoethylated once more). The ppt. was purified by chromatography over neutral Al₂O₃ or by crystn. to give 1,2,3,4-tetrakis(β -cyanoethoxy)butane (II), m. 106–8° (MeOH). 1,2,3,4-tetrakis,β-cyanoctnoxy) butane (11), m. 100-8° (MeOH). Similarly were prepd. the per(β-cyanocthyl) ethers of glycerol (ether b₀.; 234-7°), D-glucitol, D-mannitol, diglycerol, and penta-a erythritol (ether m. 48-51°). II was successfully used as the stationary phase in gas chromatography to sep. various position isomers (e.g., isomeric xylenes, cresols, Me toluates, C₁₀H₁CO₁-Me, C₂H₄(CO₁Me)₂, C₄H₄(CO₁Me)₃, Me pyridinecarboxylates, O₂NC₄H₄CN, C₆H₄(NO₁), C₆H₄(OH)₂, HOC₆H₄CO₂Me, HO-C₆H₄CN, HOC₆H₄NO₂, MeOC₆H₄NO₂, MeOC₆H₄NO₂, MeNH-C₆H₄NO₃, MeNC₆H₄NO₂) and stereoisomers (cis and trans cycloplefins and Me cycloglkanolearboxylates 2-methyl-4-ter(bytyl) olefins and Me cycloalkanolcarboxylates, 2-methyl-4-tert-butylcyclohexanols, 5-tert-butyl Me-2-hydroxycyclohexanecarboxy-lates). Finely ground unglazed tile was used as an inert support.

lates). Finely ground unglazed tile was used as an inert support.

Jiri Pliml

Synthesis and investigations of glycosylamines and attempts at glycosidation of p-aminothiophenol. Jerzy Sykulski (Akad. Med., Lodz). Acta Polon. Pharm. 20(2), 131-40(1963)(Pol). The effect of the SH group on glycosylation of amines was investigated with p-H₃NC₆H₄SH (I) derivs. as model substances. The procedure was essentially that of Weygand (CA 34, 83³).— Attempts at glycosidation of PhSH and p-AcNHC₆H₄SH, with or without HCl as catalyst, were neg. The following glycosylamines were prepd. from I [sugar component, m.p., [a]p/temp. (c1, C₈H₈N), and R_f value ascending chromatography in 2:1:3:2 BuOH-C₈H₆-C₈H₈N-H₂O given]: p-glucose, 122-5°, -120.4°/25.5°, 0.96; p-galactose, 154-6°, -123.3°/21°, 0.61; p-man-nose, 159-62°, -197.2°/21°, 0.67; L-rhamnose, 153-6°, 196.9°/21°, 0.81; lactose, 196-8°, -49.5°/19°, 0.64; maltose, 140-50°, -0, 0.72; L-arabinose, -(oil), -0, 0.64; p-arabinose, -(oil), -0, 0.68; p-ribose, -(oil), -0, 0.69; p-xylose, -(oil), -0, 0.68; p-ribose, -(oil), -0, 0.69; p-xylose, -(oil), -0, 0.68. For a glycosylamine prepd. from p-glucose and p-H₂N-C₈H₅SMe, the corresponding data were 172-5°, -0, 0.68, resp. All the compds. gave pos. color reactions for the SH group. N-(p-Mercaptophenyl)-p-glucosylamine (II) (1 g.) in 8 ml. C₈H₅Neft 3 days at room temp. with 6 ml. Ac₁O yielded 0.73 g. penta-O-Ac deriv. of II, m. 148-50°. The penta-O-Bz deriv. of II, m. 195-8°, was prepd. similarly with BzCl. II had no red ducing properties [test with o-C₆H₄(NO₁)₁], showing it to be a p-glucosylamine and not the isomeric Amadori rearrangement product. II (1 g.) dissolved in 15 ml. 5% NoOH, the soln. D-glucosylamine and not the isomeric Amadori rearrangement product. II (1 g.) dissolved in 15 ml. 5% NaOH, the soln. shaken with 7 ml. Me₂SO₄, acidified with HCl, refluxed 30 min., alkalized, extd. with Et₂O, the ext. evapd., and the residue refluxed with 3 ml. Ac₂O and poured on ice gave 0.01 g. p-AcNH-C₂H₂SMe, m. 128.5–9.7° (C₂H₂). The SH group in I was considered to have a sectorial field. sidered to have no catalytic effect on I glycosylamination.

Jerzy Lange
The synthesis of desosamine hydrochloride. A. C. Richardson (Univ. Bristol, Engl.). J. Chem. Soc. 1964(Dec.), 5364-70 e
(Eng). By use of Me 3-acetamido-4,6-O-benzylidene-3-deoxy-a-D-glucopyranoside as starting material, 3,4,6-trideoxy-3-di-methylamino-p-xylo-hexose (I) hydrochloride has been synthe-

sized; it is identical with the naturally occurring sugar desos- f

sized; it is identical with the naturally occurring sugar desos-1 amine hydrochloride.

Sorboses. V. Reactions of the p-tolylsulfonyloxy groups of 2,3-O-isopropylidene-a-L-sorbofuranoses in liquid ammonia. Kanji Tokuyama (Shionogi Co., Ltd., Osaka). Bull. Chem. Soc. Japan 37(8), 1133-7(1964)(Eng); cf. CA 61, 13394f. The reactions of p-tolylsulfonyl groups in liquid NH_s were extended to NaNH_s, KCN, KSCN, NaOH, and NaSH with the following results: in the absence of a solute 2,3:4,6-di-O-isopropylidene-1-O-p-tolylsulfonyl-a-1-sorbofuranose in liquid NH_s gave only the 1,1'-bis(amine); KCN gave an amidine which was hydrolyzed to the 1-carboxamide, [a]\(\frac{1}{2} \cdot -4.8^\text{o} \) (c 1.011, CHCl₃); g KSCN gave only the 1-amine; NaSH gave mercaptans or sulfides, depending on conditions; otherwise normal substitution occurred. Chilton H. McDonnell

Synthesis of indomethacin metabolites. R. G. Strachan, M. A. P. Meisinger, W. V. Ruyle, Ralph Hirschmann, and T. Y. Shen (Merck Sharp Dohme Labs., Rahway, N.J.). J. Med. Chem. 7(6), 799-800(1964)(Eng); cf. CA 60, 15005c. In man, the new non-steroidal antiinflammatory agent, indomethacin (I, R = OH) (II) is rapidly excreted in the urine as the acyl

glucuronide (I, R=A) (III). The chem. lability of acyl glucuronides is well known and facile hydrolytic cleavage of the glucoside linkage of III was observed during isolation. The Nglucoside linkage of III was observed during isolation. The N-aryl group was readily removed under mildly acidic or alk, conditions and in vivo diacylation gave the known acid. Preferential demethylation in C,H,N.HCl at 180° yielded the O-demethyl analog. II K salt condensed with Me 2,3,4-tri-O-acetyl-bromo-1-deoxy-a-D-glucuronate in MeOH yielded I (R = 2,3,4-tri-O-acetyl deriv. of A) (IV). Exploratory attempts to convert V into III were unsuccessful. The in vivo metabolite III (from rabbit urine) was esterified and acetylated to give IV, thus establishing the structure of III.

Synthesis of grayonoids V. Synthesis of ervodic-

5324

thus establishing the structure of III. C. R. Addinall Synthetic studies of flavonoids. V. Synthesis of eryodictiol 7-β-p-glucoside and its methyl ether. N. B. Tarusova, A. N. Vetrov, and N. A. Preobrazhenskii (M. V. Lomonosov Inst. Fine Chem. Technol., Moscow). Zh. Obshch. Khim. 34(10), 3300-3 (1964); cf. CA 58, 12500h. Reaction of I in MeOH-60% aq.

KOH at 1-2° with 3,4-(HO), C.H. CHO under N atm. gave in 1.5 KOH at 1-2° with 3,4-(HO)₂C₄H₁CHO under N atm. gave in 1.5 days 42.7% 2',3,4',6'-pentahydroxychalcone 4'β-D-glucopyranoside, m. 194-6°, which in 50% aq. MeOH heated with phosphate buffer (pH 7) 15 min. at 90° and kept 18 hrs. at 4-5° gave 3',4',5,7-tetrahydroxyflavanone 7-β-D-glucopyranoside dihydrate (II), m. 198-7°. I and isovanillin gave 52.5% 2',3,4',6'-tetrahydroxy-4-methoxychalcone 4'-β-D-glucopyranoside m. 178-80° (contg. 4.5 moles H₂O), which was converted into 72% 3',5,7-trihydroxy-4'-methoxyflavanone 7-β-D-glucopyranoside dihydrate, m. 249-50°, which in 1 hr. at 90-5° in 10% H₂SO₄ gave 98.1% 3',5,7-trihydroxy-4'-methoxyflavanone, m. 228-9°, also formed by similar treatment of neohesperidine in 77% yield. Uv and ir spectra were reported. II was identical with eryodictiol 7-glucoside.

Synthesis of coenzyme A. A. M. Michelson (Arthur Guinness

Synthesis of coenzyme A. A. M. Michelson (Arthur Guinness Son and Co. Ltd., Dublin, Ire.). Biochim. Biophys. Acta 93 (1), 71-7(1964)(Eng). An improved procedure for the chem. synthesis of coenzyme A by anhydride-anion exchange was described. A mixt. of the 2',5'- and 3',5'-diphosphates of adenosing was treated. adenosine (2',3'-cyclic phosphate)-5'-P'-diphenyl pyrophosphate quant. This was treated with pantethine 4',4'-bisphosphate in C₆H₈N to give a 2',3'-cyclic phosphate form of coenzyme A with displacement of diphenyl phosphate from the intermediate mixed anhydride. Specific enzymic opening of the cyclic phosphate gave coenzyme A in high yield. A synthesis of guanosine 2'(3'),-5'-diphosphate was also described. RCCF

Substituted 2,3-benzylidene ribonucleosides. Friedrich Cramer, Wolfram Saenger, Karl Heinz Scheit, and Juergen Tennigkeit (Tech. Hochschule, Darmstadt, Ger.). Ann. 679, 156–63(1964)(Ger). The prepn. of 2',3'-O-(4-dimethylaminobenzylidene) (I) and 2',3'-O-(2,4-dimethoxybenzylidene) ribonucleosides (II) was described. These compds. were hydrolyzed

under mild acid conditions and could be phosphorylated with 2-cyanoethyl phosphate (III) (Tener, CA 55, 24771g) and dicyclohexylcarbodiimide (DCC). After cleavage of the protec-2-cyanoethyl phosphate (III) (Tener, CA 55, 24771g) and dicyclohexylcarbodiimide (DCC). After cleavage of the protective groups, nucleoside 5'-phosphates were obtained exclusively. Uridine (2 millimoles), 5 millimoles 4-Me₁NC₄H₄CHO, 4 cc. dioxane, and 18 millimoles CF₂CO₂H (IV) or CCl₂CO₃H (V) heated slightly in a closed vessel, and the homogeneous liquid shaken 24 hrs. at room temp. and worked up gave 67% 2',3'-O-(4-dimethylaminobenzylidene)uridine, m. 203-4° (H₂O contg. a little EtOH). From cytidine was similarly prepd. 78% 2',3'-O-(4-dimethylaminobenzylidene)cytidine (VI), m. 209-10° (H₂O). From adenosine was prepd. 69% 2',3'-O-(4-dimethylaminobenzylidene)adenosine, m. 218-20° (H₂O contg. a little EtOH). With guanosine, the reaction did not proceed to completion work-up gave 0.4 g. 2',3'-O-(4-dimethylaminobenzylidene)guanosine, m. 269-70°. VI (2.5 millimoles) in 10 cc. HCONMe₂ (DMF) was treated as described for N*-benzoyl-2',3'-O-(4-dimethylaminobenzylidene)cytidine (VII) (see below) and worked up (chromatography) to give 69% N*-benzoyl-2',3'-O-(4-dimethylaminobenzylidene)cytidine (VIII), m. 173-5° (CHCl₁-petr. ether) (McOH); sometimes VIII crystd. as VIII.H₁O, m. 124-6°. In

Survey 4302D

give the base hydrochloride, which was hygroscopic and crystd. from abs. EtOH-ether (yield 20%). Following II were prepd. (R, n, and m.p. or b.p. given): Et, 2, b, 200-2°; Pr, 3, m. 190-1° (softened at 165°) (hygroscopic, isolated as hydrochloride); Bu, 2, b, 222-5°; and Et, 3, b, a 210-14°. I (1 mole) in dry C₆H₆ or xylene (20 ml.) was treated with the appropriate alkyl or aryl halide (1 mole) and the mixt. refluxed (4-24 hrs.) to give the quaternary salt (III); this was filtered off, washed (dry C₆H₆) and dried in vacuo. The following III were prepd. (R, R¹, X, n, and m.p. given): Me, allyl, Br, 2, 225-7°; Me, Me, I, 2, 270° (decompn.); Me, Et, I, 2, 228-9° (decompn.); Me, Pr, I, 2, 178-80°; Me, benzyl, Br, 2, 148-9° (sepd. with 2H₆O); Me, Bu, I, 2, 160-1°; Me, p-nitrobenzyl, Br, 2, 227-9° b (decompn.); Et, benzyl, Br, 2, 202-3°; Et, Me, I, 2, 152-5° (decompn.); Et, benzyl, Br, 2, hygroscopic; Pr, Me, I, 2, 170-2°; Pr, benzyl, Br, 2, 148-50°; Pr, p-nitrobenzyl, Br, 2, 204-5° (decompn.); Bu, Me, I, 2, 160° (decompn.) (softened at 143°); iso-Bu, Me, I, 2, 160° (decompn.) (softened at 143°); iso-Bu, Me, I, 2, 154-6°; Ht, Me, I, 3, 133-5°; Et, Et, I, 3, 111-12°; Et, benzyl, Br, 3, hygroscopic; Bt, p-nitrobenzyl, Br, 3, 205-6°; Bu, benzyl, Br, 3, hygroscopic; and Bu, p-nitrophenyl, Br, 3, 78-9°. Some of the compds. reported possessed pronounced spasmolytic activity.

V. K. Ahluwalia

2,5-Dimethylpiperazine synthesis from 1-amino-2-propanol. W. K. Langdon, W. W. Levis, Jr., and D. R. Jackson (Wyandotte Chemicals Corp., Wyandotte, Mich.). Ind. Eng. Chem. 1, No. 2, 153-6(1962). In the presence of Raney Ni catalyst (I) and H, 1-amino-2-propanol (II) undergoes a bimol. cycloamination yielding 2,5-dimethylpyrazine (III), trans-2,5-dimethylpiperazine (IV), and cis-2,5-dimethylpiperazine (V). I and II are charged into a 1-1. stainless steel autoclave in such amts. that there are 2.5 g. of I per mole of II. The autoclave is then flushed several times with N and then with H. H is admitted to 1200 lb./sq. in. and the mixt. rapidly heated to 220° venting H to maintain 1200 lb./sq. in. The mixt. is agitated 4-8 hrs., cooled to 80°, the H released, and 200 ml. H₂O added. I is removed by filtration through Celite and the filter cake washed with H₂O. V is removed as an azeotrope with water, b. 98°, through a 2.8 × 120 cm. packed column with a reflux ratio between 2:1 and 5:1. IV and V were distd. at 155-70°. There is a 75% yield of a mixt. of 80% IV and 20% V. The following changes in reaction variables all resulted in greater yields of IV and V: increasing temp. from 180 to 220°, increasing H pressure from 200 to 1200 lb./sq. in., increasing reaction time from 2 to 8 hrs., and increasing g. of I per mole of II from 0.6 to 2.5. Thomas E. Whiteley Condensation reaction of amino alcohols with imides.

IV. Condensation products of β-(cycloamino)ethanols with—succinimide and with phthalimide—formation of β-(cycloamino)ethylamines. Kazuo Nakajima (Teikoku Chem. Ind., Mibukayogosho, Kyoto). Nippon Kagaku Zasshi 81, 1129–32(1960); cf. CA 57, 8545h. Seven aminoales. (RNCH₂CH₂OH) (I), where RN = (CH₂)₄N (a), (CH₃)₆N (b), β-tetrahydroquinoline (c), morpholino (d), 4-ethylpiperazino (g). I and phthalimide heated at 200° while removing the H₂O formed yielded RNCH₂CH₂N(CO)₂C₆H₄-o (II) (RN, m.p., b₁₀, m.p. of picrate, and decompn. point of Pt salt given): a, 109°, 220°, 218°, 212°; b, 90°, 206°, 215°, c, 137°, 273°, 144°, 204°; d, 135°, 216°, 232°, 146°; e, 115°, 227°, 245° (decompn.), 265°; f, 75°, 252°, 260° (decompn.), 271°; g, 158°, 279°, 218° (decompn.), 262°. h Similarly, I and succinimide yielded RNCH₂CH₂N(COCH₂)₂ (III) (RN, m.p., b₁₀, m.p. of picrate, and decompn. point of Pt salt given): a, 55°, 170°, 158°, 212°; b, —, 170°, 190°, 230°; c, 115°, 210°, 180°, —; d, 75°, 170°, 181°, 175°; e, —, 199°, 245°, 245°; f, 65°, 223°, 225°, 255°; g, 141°, 272°, 175°, 236°. II and III refluxed with 20% HCl 3–5 hrs., phthalic acid or succinic acid filtered off, and the filtrate concd. yielded RNCH₂CH₂NH₂HCl (IV.HCl). Free RNCH₂CH₃NH₂(IV) was obtained by a long extn. of neutralized IV.HCl with ether. (RN, b₁₀, b₁₅₀, m.p. of picrate, and decompn. point of Pt salt given) a, 33°, 165–6°, 210° (decompn.), 207°; b, 50°, 185°, ~200 (decompn.), 218°; c, 157°, 290°, —, —; d, 81°, 204°, 177° (decompn.), 237°;

e, 88°, 223-4°, 230°, 270°; f, 120°, 255°, 116°, 252°; g, (m. 80°), 191°, 326°, —, 210-40°. H. Kubo Synthesis of potential anticancer agents. XXX. (1-Aziridinyl)purines. J. A. Montgomery, K. Hewson, and C. Temple, Jr. (Southern Res. Inst., Birmingham, Ala.). J. Med. Pharm. Chem. 5, 15-24(1962); cf. CA 56, 1445a. Eight (1-aziridinyl) purines were prepd. by nucleophilic displacement of the Cl in the corresponding chloropurines. The compds. were screened against Ca 755 and Walker 256, and the 6-(1-aziridinyl)purines show moderate activity against the Ca 755, and in 1 case good activity against Walker 256. 6-(1-Aziridinyl)-9-benzyl-9H-purine (1) (100 mg.) in 10 ml. EtOH was hydrogenated at room temp./atm. pressure with 50 mg. 5% Pd-C catalyst 18 hrs., the catalyst filtered off, the filtrate evapd. to dryness, the residue dissolved in 10 ml. 0.1N HCl, the soln. evapd. to dryness in vacuo, the residue triturated with 10 ml. Me₂CO, the ppt. filtered off, and dried in vacuo to give 90 mg. 9-benzyl-No-ethyladenine-HCl. I can also be reduced with metallic Na in liquid NH_a to give N⁶-ethyladenine, m. 227-8° (decompn.). A mixt of 5.7 g. 2,6-dichloropurine, 6.9 ml. benzyl chloride, 4.2 g. K₂CO₃, and 50 ml. Me₂SO was stirred 18 hrs., filtered, the filtrate concd. to one-quarter vol. in vacuo, poured into ice-H₂O, the oil allowed to solidify, filtered, washed with H₃O, air dried, and recrystd. from 75% aq. EtOH with Norit treatment to give 5.9 g. of a mixt. of the 7- and 9-isomers of benzyl-2,6-dichloro-9*H*-purine, m. 122-30°. Recrystn. from MeOH gave 1.06 g. pure 9-isomer, m. 148°. The mother liquors from the recrystn. gave 30% yield of the 7-isomer, contaminated with the 9-isomer. Prepd. similarly was 2-amino-9-benzyl-6-chloro-9*H*-purine, and 2-amino-7-benzyl-6-chloro-7*H*-purine. 9-Benzyl-2,6-di-chloropurine (278 mg.) in 10 ml. 0.2*N* NaOH was refluxed with stirring to effect soln., the hot mixt. filtered, the filtrate acidified with glacial HOAc, the ppt. filtered off, washed with dil. HOAc, and dried in vacuo to give 233 mg. 9-benzyl-2-chlorohypoxanthine (II), m. 245°. II (100 mg.) in 15 ml. abs. EtOH was placed in a glass-lined Parr bomb, satd. with dry NH, at 5°, the soln. heated 6 hrs. at 150°, the bomb chilled, opened, the soln. concd. to 10 ml. in a stream of dry N, the solid filtered off, washed with EtOH, and dried in vacuo to give 53 mg. 9-benzylguanine (III), m. 303°. III may also be prepd. from 2-amino-9-benzyl-6-chloropurine by reaction with HCl. II (0.29 millimole) and 0.2 ml. 25% aq. dimethylamine were dissolved in 10 ml. dioxane, the mixt. heated 20 hrs. at 95-100°, with an addn. of another quantity of Me₂NH at the end of the first 2 hrs., the mixt. evapd. to dryness in vacuo, the residue triturated with Et2O, and the residue recrystd. from BtOH to give 57 mg. 9-benzyl-N², N²-dimethylguanine, m. 282°. 7-Benzyl-2,6-dichloro-purine (contaminated with the 9-isomer) (1.11 g.) in 25 ml. N NaOH was heated 1.5 hrs. at 115°, cooled to room temp., kept overnight in the refrigerator, the solid filtered off, dis-solved in H₂O, acidified with glacial HOAc, the solid filtered off, dried in vacuo, and recrystd. from abs. EtOH to give 60 mg. 7-benzyl-2-chlorohypoxanthine, m. 285°. 6,8-Dichloropurine (1.06 g.), 0.60 g. anhyd. K₂CO₃, and 1.4 ml. PhCH₂Cl in 10 ml. HCONMe₂ was heated 1 hr. at 70-85° with stirring, concd. to 0.5 vol. in vacuo, poured into ice-H₂O, the oil extd. with Et₂O, the ext. dild. with EtOH, the solid filtered off, washed with Et2O, dried, extd. with Skellysolve C, the ext. evapd. to an oil, dissolved in 1:1 aq. EtOH, the solid filtered off, washed with H2O, dried in vacuo over P2O6 at room temp., and recrystd. from aq. EtOH to give 203 mg. 9-benzyl-6,8-dichoro-9H-purine, m. 92°. Prepd. by similar methods was 9-benzyl-6-dimethylamino-9H-purine, m. 117°. 9-Benzyl-6-dimethylamino-8-chloropurine (307 mg.), 43 mg. MgO, and 34 mg. 5% Pd-C in 30 ml. abs. EtOH was hydrogenated at atm. pressure 6 hrs., the catalyst filtered off, the filtrate evapd. to dryness, and the residue triturated with EtOH to give 135 mg. 9-benzyl-6-dimethylamino-purine, m. 128°.

Aletha Kowitz

See also: Identification and detn. of o-phenylenediamine in residual waters, in the synthesis of 2-sulfanilamido-quinoxaline (sulfaquinoxaline), by paper chromatography (Sterescu) 24. Mechanism of the aminonitrile rearrangement (Ioffe) 27. Reaction of 2,3-dihydrofurans with 2,4-dinitrophenylhydrazine (Boberg) 31. Kinetics of the reversible hydration of 2-hydroxypteridine (Inoue) 5. Nucleic acid components and their analogs—synthesis of 3-methyl-6-azauridine 5'-phosphate and pyrophosphate (Zemlicka) 37.

Univ.). Vitamins (Kyoto) 10, 353-9(1956).—S-Acetylthiamine diphosphate (I) was synthesized by the acetylation of cocarboxylase in alk. medium with Ac₂O. The S-acetyl group in this compd. underwent transacetylation with cysteine as well as glutathione in aq. media. The hydroly-a sis of I in the presence of chicken liver or pigeon breast muscle homogenate was greatly promoted by the addn. of glutathione.

Reduction of the aci-reductone hydroxycomenic acid; peroxidative effects of yeast nucleic acid and adenine. Hans v. Buler and Hans Hasselquist (Univ. Stockholm). Ann. 604, 41-7(1957).—Meconic acid (1 g.) in 10 cc. 6% HBr soln. heated 3 hrs. on a steam bath, the mixt. cooled to 0° and treated carefully with 0.3 cc. Br, the suspension of bromocomenic acid heated on a steam bath, the cooled soln. extd. with EtOAc, the dried ext. concd., and the cryst. product recrystd. from EtOAc gave hydroxycomenic acid (I), m. 243°, reducing Tillmans reagent, iodine soln., and giving a red-violet ppt. with 2,3,5-triphenyltetrazolium chloride; Et ester, reducing Tillmans reagent. I (0.3 g.) in 5 cc. H₂O shaken with 11.4 cc. soln. (4.5 g. Br in 100 cc. CHCl₂), the aq. layer washed with CHCl₂ and extd. with EtOAc, the filtered ext. dried and evapd., and the residue crystd. from EtOAc gave 80 mg. brown prisms of a bromo compd., C₈H₈BrO₈, purified by recrystn. to a pure material, m. 120° [hardening and melting sharply at 170° (decompn.)]. In connection with the effect of I in combination with chemotherapeutic agents on rat sarcoma, in vitro expts. on per—oxidase systems were reported. Pyrogallol (15 mg.), 0.5 cc. 6% H₂O₂, and 5 cc. buffer soln. (pH 6.8) at 20° in 30 min. gave a min. amt. of purpurogallin (II) (E_{1 cm.} 0.002). Addn. of 30, 15, 7.5 mg. yeast nucleic acid per cc. of above reaction mixt. gave 2.12, 1.02, 0.66 mg. II (E_{1 cm.} 0.06, 0.03, d 0.017) corresponding to 7.1, 6.8, 8.8 mg. II per 100 mg. nucleic acid. Addn. of 5.0, 2.5, 1.25 mg. adenine to the mixt. gave 64.2, 64.4, 67.1 mg. II per 100 mg. adenine. Test of cleavage products of yeast nucleic acid and other purine and pyrimidine bases showed only adenosine and adenosine phosphoric acid to have equiv. activity to ade-In connection with the effect of I in combination with chemoadenosine phosphoric acid to have equiv. activity to adenine. Guanine and uracil have much less activity. Of the amino acids only p-H2NC6H4CO2H was significantly active. Xanthine, clupeine, theophylline, tyrosine, and suprarenin e proved inactive. Pyrogallol (150 mg.) in 30 cc. H₂O₂, 10 cc. phosphate buffer (pH 6.8), 1 cc. 6% H₂O₃, and 1 cc. yeast emulsion warmed 10 min. at 20, 50, 56, 60, and 69 gave relative peroxidase effects of 6.9, 8.0, 18.2, 10.2, and 0.7, confirming previous findings (cf. E. and Blix, C.A. 13, 3210) that the effect of catalase is increased 3-4 fold by warming to 56°.

C. R. Addinall

Chemotherapeutic studies in the heterocyclic series. XVIII. Polymethylenedi(sydnones) and polymethylenedi-f (hydrazines). H. U. Daeniker and J. Druey (Ciba A.-G., Basel, Switz.). Helv. Chim. Acta 40, 918-32(1957), cf. C.A. 51, 10518b.—A mixt. of [(CH₉)_nNH₂]_s, CH₂O, and HCN or KCN forms with HNO₂ followed by boiling with KOH the following [(CH₉)_nN(NO)CH₂CO₂H]_s (I) (n and m.p. given): 1, 143.5-5.0°; 2, 142°; 3, 126-7°. I and Ac₂O

or (CF₁CO)₂O form the di(sydnones) (II) [(CH₂)_nN.-

N.O.CO.CR]₂ (R = H) which can be halogenated with a gnydrohalic acid and halate to II (R = Cl or Br). The following II are reported (n, R, and m.p. given): 1, H, 169°; 1, Cl, 120-30° (explodes); 1, Br, 163° (explodes); 2, H, 170°; 3, H, 114-16°; 3, Br, 120-2°. II (n = 1, R = Me),—decomp. 182-3°, \(\) 300 m\(\mu \) (e 15,800), is prepd. using AcH via [CH₂N(NO)CHMeCO₂H]₂, m. 164-5°. II with HCl in MeOH form the following [(CH₂)_nNHNH₂.HCl)]₂ (III) (n and m.p. given): 1, 160°; 2, 156-60°; 3, 139-41°. III (n = 3) and BzH form the dihydrazone (IV), m. 80°, but III (n = 1 or 2) form compds. m. 132-3° and 154-5°, resp., in which a third benzylidene group bridges the 2 central N atoms. These compds. lack any absorption in the 3 \(\mu \) region characteristic of NH vibrations. IV absorbs as expected at 2.93 \(\mu \). III (n = 1) and 3 moles \(\nu \)-HOC₂H₄CHO form an analogous cyclic dihydrazone, m. 190°. III and \(\nu \)-C₂H₄(CHO)₂ form 2,2'-polymethylenebis(phthalazinium salts) [(CH₂)_nC₂H₃N₂Y]₂ (V). V (n = 1, Y = Br) was also prepd. from phthalazine and (CH₂Br)₂. The following V are reported [n, Y, m.p., and \(\lambda \) in \(\mu \) (e) given]: 1, Cl, 221°, 311 (7800), 282 (7600), 233 (41,000); 1, Br, 277-8°, 310 (8400), 280 (8050), 234 (52,400); 2, Cl, 285-6°, 318 (7200), 278 (6800), 234 (51,000); 3, Cl, 248°, 318 (6900),

278 (6500), 234 (60,000). V (n=2, Y = Cl) and alkali form an impure 2,2'-tetramethylenedi(1-hydroxy-1,2-dihydrophthalazine), m. 164°. V (n=1) forms instead the cyclic anhydride of the analogous compd., m. 150-2°, λ 308 (17,800), 234 m μ (17,750), 6.22, 6.47, and 6.73 μ but no OH band in the 3 μ region. 1-Hydroxy-2-butyl-1,2-dihydrophthalazine, prepd. for comparison, m. 67-8°, λ 310 (8850), 234 m μ (15,500), 2.81 μ . Acidification of all of these compds. reforms V. N-Butylphthalazinium chloride m. 64-84°, λ 318 (3200), 277 (3200), and 234 m μ (36,000). II and III (n=1, n=1, n=1 or halogen) weakly inhibit the in vivo formation of tumors and are nontoxic. J. H. S.

Structure of proteins. IV. Progressive degradation of the polypeptide chain at the N-terminal amino acid. I. Formation of 7-nitro-3,4-dihydro-2-hydroxyquinoxalines. Ernesto Scoffone, Elio Vianello, and Alberto Lorenzini (Univ. Padua). Gazz. chim. ital. 87, 354-65(1957); cf. C.A. 51, 13955a.—An improved method of identification of amino acids is given. The 2,4-dinitrophenyl deriv. is suspended in 6 ml. 95% EtOH and heated with 6 ml. concd. (NH₄)₈S 30 min. at 45-50° to give the 2-amino-4-nitrophenyl deriv. This is cyclized by acidification to pH 2-3 and heating 15 min. at 70° to 7-nitro-3,4-dihydro-2-hydroxy-quinoxaline (I) or its 3-substitution product. The method is preferable to that of complete reduction with Pt-H to the 7-amino compds. (Jutisz and Ritschard, C.A. 50, 4052i), which are sensitive to air-oxidation. The m.ps. of the dinitrophenyl derivs. are: glycine, 191-1.5° (decompn.); alanine, 173-4°; valine, 182-3°; leucine, 131-2°; serine, 186-8°; threonine, 176-7°; aspartic acid, 187-8° (decompn.); glutamic acid, 158-62°; methionine, 108-10°; phenylalanine, 208-10° (decompn.) of I and its derivs. are: I, 254-4.5°; 3-Me, 224-5°; 3-iso-Pr, 221.5-2.6°; 3-iso-Bu, 217-18°; 3-HOCH₂, 221-2°; 3-MeCHOH, 227-7.5°; 3-MeSCH₂CH₂, 213-14°; 3-Bz, 210-11°; 3-(3-indolylmethyl), 220-1°. Data on ultraviolet absorption max. are given, as are R, values for chromatographic sepn. I is oxidized by alk. KMnO₄ to 7-nitro-2-hydroxyquinoxaline (II), m. 268-9°, and by K₃Cr₂O₇ to 7-nitro-2,3-dihydroxyquinoxaline (III), m. 363-4°. I, II, and III were identified by independent synthesis from 4-nitro-o-phenylenediamine and HOCH₃CO₂H, (RO)₂CHCO₂H, and (CO₂H)₂, resp.

Triethylenediamine (1,4-diazabicyclo[2,2,2]octane) and hexaethylenetetramine. III. Interaction of 2,2',2''-trichlorotriethylamine hydrochloride and dimethylamine. Frederick G. Mann and F. C. Baker (Univ. Cambridge, Engl.). J. Chem. Soc. 1957, 1881-99; cf. C.A. 50, 358h; Hromatka and Kraupp, C.A. 46, 8123i.—When a mixt. of (ClCH₂CH₂)₁N.HCl (I) and a 40% soln. (3.4 moles) of methanolic Me₂NH (II) is set aside at room temp. 24 hrs. or heated at 45-50° 1.5 hrs. and then treated with HCl, 1-(2'-chloroethyl)-4,4-dimethylpiperazinium chloride hydrochloride monohydrate (III) is isolated. III is also obtained when the proportion of II is varied from 2.0 to 6.6 moles with the same heating. III, hydrated or anhyd., m. 264° with two well-defined stages of effervescence. An MeOH soln. of III when shaken with Ag₂O and treated with MeI gives the piperazinium iodide in the cold and the iodide methiodide when refluxed. When I is heated with methanolic II (12.8 moles) at 40-5° 8 hrs., the product is 1-(2-'dimethylaminoethyl)-4,4-dimethylpiperazinium chloride-2HCl (IV), m. 227-8°. When heated with methanolic II (10.7 moles) at 125° 7 hrs., I yields 1-(2'-dimethylaminoethyl)-4-methylpiperazine-3HCl (V), m. 262° after recrystn., 270° after sublimation. The free base (Va), b₁₂ 94°, is liberated by treating V with concd. aq. alkalies. III gave triethylenediamine-HCl methochloride on heating at 250-4°. Due to different results obtained by other workers, the prepn., properties, and identification of triethylenediamine (VI), m. 156-7°, were studied and a no. of derivs. of this and other amines were prepd. The thermal decompn. of IV and V at atm. pressure gives similar products. Thus, using V, distn. of the ethereal: exts. of the liberated amines gives the following fractions (1) the ether distillate, contg. 1,4-dimethylpiperazine (VII), b. 131-2°; (2) a fraction, b. 130-3°, of VII contg. a small amt. of piperazine; (3) a large fraction, b₁₂ 86-7°, consisting

of Me₃NCH₂CH₂N.(CH₂)₂.NMe.(CH₂)₂ (VIII), contaminated with a base, $C_6H_{19}N_3$; (4) a small fraction, $b_0...91^\circ$, m. 40–3°, which was shown by independent synthesis to be 1,2-

CADLT ONSMER

bis (4-methyl-1-piperazinyl) ethane (IX), and (5) a small amt. of bis(4-methyl-1-piperasinyl)ethane(IX), and (b) a small amt. of a fraction, b_{0.061} 105-8°, believed to be 1,4-bis[2-(4-methyl-1-piperazinyl)ethyl]piperasine. The 4,4'-di-Me deriv. of IX was prepd. by hydrolysis of the 4,4'-bis(ethoxycarbonyl) deriv. followed by H₂CO-HCO₂H methylation of the hydrolysis a product. The products of the thermal decompn. of IV differ from those of V only in that fraction 2 contained only a trace from the second of the hydrolysis and the second only a trace of the second only a trace of the second of the hydrolysis and the second only a trace of the second of the second only a trace of the second only a of piperazine, fraction 3 is almost pure VIII, and fraction 5 is not detected. Possible errors in earlier literature in the identification of certain of these amines is discussed and the usefulness of the table of derivs. below to differentiate among these compds. is emphasized. The following derivs. of VI, IX, these compds. is emphasized. The following derivs. of VI, IX, VII, and Va were prepd. (m.p. given): hydrochloride, 320°, 286-8°, 265°, 262°; hydrobromide, 360°, 267°, 244°, 239°; hydriodide, above 360° (chars without melting), 231° (tri salt), 237°, 266° (di salt); methobromide, 296°, 296° (tri salt), 364° (evacuated tube) 315° (di salt); methiodide, 284°, 288° (di salt) (tri salt, m. 270°), 334° (sealed tube), 320° (di salt); methopicrate, 324°, 212° (tri salt), 320°, 220° (di salt); p-nitrophenoxide, 182-3°, 184°, 125-6°, 139°; 2,4-dinitrophenoxide, 158°, 221°, 216°, 165°; picrate, 290°, 292°, 300°, 254°; nitrate, 176°, 245°, 210°, 164°; oxalate, 290°, 238-40°, 276°, 228°; and chloroplatinate, 324°, 287°, 282°, 242°. VI was studied against an infection of Heterakis spumosa in mice. It shows no activity at doses of 1.25 g./kg. spumosa in mice. It shows no activity at doses of 1.25 g./kg. given three times. Piperazine is fully active at this dose VI produces no marked pharmacol. effects in the cat, except transient decrease in blood pressure. VI is also without action in cotton rats infected with Litomosoides carinii at a dose of 1 g./kg. IV and V are inactive as possible schistosomicides. Robert Filler

Electrophilic substitution X. Nitration of quinoxaline. M. J. S. Dewar and P. M. Maitlis (Univ. London). J. Chem. Soc. 1957, 2518-21; C.A. 51, 12091g.—Quinoxaline (I) was shown to be mononitrated in the 5-position (Π) and dinitrated in the 5,6-positions (III). I (11 g.) in 50 ml. H₂SO₄ was added to a mixt. of 25 ml. HNO₄, 50 ml. H₂SO₄. and 50 ml. 65% oleum, the mixt. heated 24 hrs. at 85-90°, cooled, and hydrolyzed by pouring on ice, the solid extd. with CHCl₂-Me₂CO, and the ext. evapd. gave 12 g. tar, which purified gave 5.1 g. III, prisms, m. 172-3°. Evapn. of the mother liquors and chromatography of the residue e gave 0.2 g. more III and 0.3 g. II, m. 95-6° as yellow needles. Only 0.4 g. of I was recovered by neutralizing the origina soln. N₂H₄.H₂O (3 ml.) and 50 mg. Pd-C added at 70° to soln. N₂H₄.H₂O (3 ml.) and 50 mg. Pd-C added at 70° to 0.9 g. III in alc., after 0.5 hr. the soln. filtered, the filtrate evapd., and the product warmed 5 min. with 10 m. Ac₂O and 0.1 ml. concd. H₂SO₄ gave 0.1 g. 5-acetamidoquinoxaline, m. 130-1°. III (0.95 g.) refluxed 0.5 hr. in AcOH with 1 g. Fe powder, and the product hydrolyzed, neutralized, and acetylated gave 0.25 g. 2'-methylglyoxalino[4',5':5,6]-quin-foxaline (IV), m. 249-9.5°. The formation of IV suggested that the 2 original NO₂ groups were ortho to each other. The reduction was then repeated and the products sepd. at the amine stage by chromatography into a small amt. of IV and 0.2 g. diamine (V). Phenanthraquinone (0.15 g.) in 1 and 0.2 g. diamine (V). Phenanthraquinone (0.15 g.) in 1—ml. AcOH warmed with 0.14 g. V in 1 ml. alc. gave 0.2 g. 1,2,3,4-dibenzopyrazino[2',3':6,7]phenazine (VI), m. above 300° (from MeNO₂). When the spectrum of VI compared with that of 1,2,3,4,5,6-tribenzanthrene, allowing for the difference in intensity of the various peaks, a bathochromic shift of shout 12 ml for the hydrogenear and the contact of the difference in intensity of the various peaks, a bathochromic shift of about 13 m μ for the hydrocarbon made the spectra superimposable. Fe powder (0.5 g.) refluxed 0.5 hr. with 0.5 g. III, 3 ml. Ac $_2$ O, and 2 ml. AcOH, and the product chromatographed on Al $_2$ O, gave 0.45 g. 5,6-diacetamido-quinoxaline (VII), m. 232-3° (from aq. alc.). Attempts to hydrolyze VII failed. After heating 0.5 hr. with 5N HCI unchanged VII was recovered. Hydrolysis by 5N NaOH gave tars. I (1 g.) in 12 ml. (CF₂CO)₂O and 0.5 ml. HNO₄ gave tars. I (1 g.) in 12 ml. (CF₆CO)₂O and 0.5 ml. HNO₆ left 88 hrs. at room temp, gave 0.43 g. hydroxynitroquinoxaline, m. 182–3° (from Me₂CO), which did not correspond to any of the known isomers. XI. Nitration of some sixmembered N-heterocyclic compounds in sulfuric acid. *Ibid.* 2521–8.—The proportions of mono-NO₅ derivs. formed from quinoline (I) and isoquinoline (II) in H₅SO₄ were detd. I gave at 0° 47.7% 5- (III) and 52.3% 8-nitroquinoline (IV); II gave 5- (V) and 8-nitroisoquinoline (VI) in the proportions: 90.4% V, 9.6% VI at 0°; 84.8% V and 15.2% VI at 100°. Both reactions took place readily under mild conditions. For II a value for the parameter α was derived from the partial rate factors and was used, together with a value the partial rate factors and was used, together with a value for β , for predictions of the reactivities of different positions in several 6-membered heterocyclic systems, giving excellent

agreement with expt. Competitive nitrations were also carried out with I-II and I-acridine (VII); their significance was discussed. The relative reactivities were: I:II = :24.5; I:VII = 1:190. Attempts to follow the kinetics failed because the reactions were essentially complete in less than 1 min. at 0°. The π -electron energy difference was calcd. from the equation: $\Delta E_{\pi} = 2\beta_{\pi}$ (a_{or} + a_{or}) where atoms r and s were those adjacent to the point of attack. The value for β_x was found to be -6 kcal./mole for nitration in Ac_2O and -4 kcal./mole for nitration in H_2SO_4 . Thus it was shown that for a given reagent X, the velocity constant K, was given by $\log K_r = A_x - (\Delta E_{\pi})_i/RT$. Thus $\alpha = 39.7 \pm 2$ kcal./mole when calcd. from the above equations for VI. The theory broke down in only 2 cases; 1st, when the position being considered was ortho to the ring N atom (as the 8-position in I and the 1-position in VII); 2nd, for positions analogous to the 4-phenanthryl positions. The results for a series of calcus, on the following 6-membered heterocycles were listed: pyridine, I, II, quinoxaline, quinazoline, cinnoline, VII, phenanthridine, benzo[f] quinoline, phenazine, and benzo[c]cinnoline. From results the ratio of the partial rate factors Ky: KIII was calcd. to be 46.4:1 and $\Delta E_{\pi}(III) - \Delta E_{\pi}(\nabla)$ was found to be -1.3 kcal./mole. This gave a theoretical value for the ratio of these partial rate factors of 10.5:1. HNO₁ (6.2 ml.) heated 1 hr. at 95° with 18 g. isoquinoline N-oxide in 80 ml. H₂SO₄ gave 13 g. 5-nitroisoquinoline N-oxide, m. 220-1° (from MeNO₂), and ochromatography of the mother liquors yielded 2.2 g. iso-quinoline N-oxide and 1.4 g. 8-nitroisoquinoline N-oxide (VIII), m. 188-9°. VIII (0.5 g.) refluxed 0.5 hr. in 7 ml. PCl₃ gave 65% VI, m. 87-7.5°. Reduction of VI with N₂H₄.H₃O and Pd-C gave 60% 8-aminoisoquinoline, yellow needles, m. 170-1° (from ligroine). V was prepd. from the N-oxide in an analogous manner in 52% yield. I and II were discoursed in M-SO. a solve of HNO added gradually and discoursed in M-SO. dissolved in H₅SO₄, a soln. of HNO₂ added gradually and after 0.5 hr. the solns. poured on ice. The mononitro fraction was sepd. by partial basification (at pH 2.1 for I; at pH 2.5 for II) from unchanged base and analyzed spectroscopically. With I, both III and IV had peaks around 300 $m\mu$ and troughs at 250 m μ , whereas the 3-, 6-, and 7-isomer had considerable peaks at 250 m μ . B. K. Wasson

16492

and troughs at 250 mµ, whereas the 3-, 6-, and 7-isomer had considerable peaks at 250 mµ.

Syntheses with bis(pyridinium salts); a new pathway to the phenazine series. Otto Westphal and Klaus Jann (A. Wander-Forschungsinst. Freiburg/Zähringen, Ger.). Ann. 605, 8-15(1957).—(CICH3CH:)2(18 g.) in 50 cc. PhCH2OH and 40 cc. pyridine kept 3 days at 20° and treated with 50 cc. Me3CO gave 32 g. ((ChH3CH2CH:)2(I), hygroscopic, m. 209-10°, 2.83 g. of which in 16 cc. 50% EtOH with 3.3 g. p-ONC4H4NMe2 (II) in 35 cc. EtOH and 4.2 g. piperidine after several hrs. at 0° gave 3.05 g. [p-Me2NC4H4NO:)CHCH:]2 (fumaraldehyde dinitrone), red microneedles and stars, m. 238° (decompn.) (from hot PhNO2). I (1.5 g.) in 10 cc. 50% EtOH with 1 g. II in 20 cc. EtOH and 0.5 g. NaCN in 5 cc. H20 at 20° gave 0.8 g. of the fumaraldehyde bis(cyanoanil), [p-Me2NC6H4N:C(CN)CH:)2, dark red needles, m. 330° (decompn.) (after washing with 50% EtOH), which were not recrystd. (BrCH2CO)2 (10.7 g.) in 40 cc. EtOH at 0° with 6 g. 1,2-Ch4(NH2)2 gave 12 g. 2,3-bis(bromomethyl)quinoxaline (III), yellow needles, m. 150-1° (from EtOH). III (10 g.) in 50 cc. hot Me2CO and 40 cc. EtOH added to 12 cc. pyridine gave 11.4 g. quinoxaline 2,3-bis(methylenepyridinium bromide) (IV), C2H18NB72, hexagons, m. 204-5° (from abs. EtOH); dihydrate, m. 264-5° (from aq. EtOH). IV (5 g.) in 30 cc. 50% EtOH treated with 3.8 g. II in 45 cc. EtOH, cooled to 10°, and 2.2 g. NaCN in 8 cc. H2O added after further cooling gave almost quantitatively the corresponding bis(cyanoanii) of 2,3-quinoxaline dialdehyde. C2H2NN. red needles with coppery quantitatively the corresponding bis(cyanoanil) of 2,3quantizative dialdehyde, C₁₈H₂₄N₈, red needles with coppery surface luster, decomp. about 226° (from pyridine). IV (1 g.) and 0.7 g. Ac₂ in 10 cc. EtOH at 0° treated with 0.9 cc. Bu₂NH and kept 2 days at 0° gave 570 mg. crude *Jamino-2,3-dimethylphenasino-4-pyridinium bromide dihydrate* amino-2,3-dimethylphenasine-4-pyridinium bromide dihydrate (V), C₁₈H₁₇N₁Br.2H₂O, red microneedles, m. 304-5° (from H₂O), 400 mg. of which were refluxed 4 hrs. with 2.5 cc. pyridine, and, after 12 hrs. at 20°, washed with H₂O, dried, and sublimed at 180°/0.002, to give 200 mg. 1,4-diamino-2,3-dimethylphenasine (VI), diamagnetic, dark blue-violet needles, decomp. about 215°, and giving a blue EtOH soln. which was decolorized by AcOAg with *sepn. of Ag. VI (70 mg.) heated 5 min. with 3 cc. pyridine and 2 cc. Ac₂O gave almost quantitatively the 1,4-di-Ac deriv. (VII) of VI, lemon-yellow needles, m. 333-4° (after sublimation at 270°/760). V (0.5 g.) refluxed 5 min. with 5 cc. pyridine